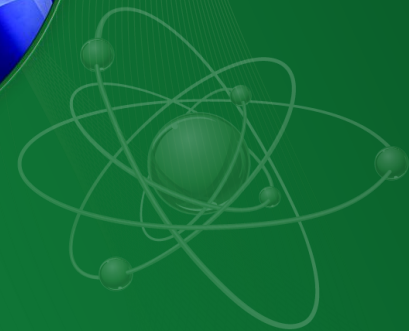
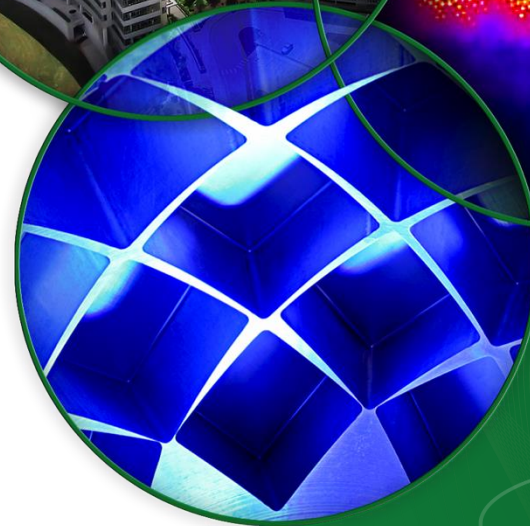
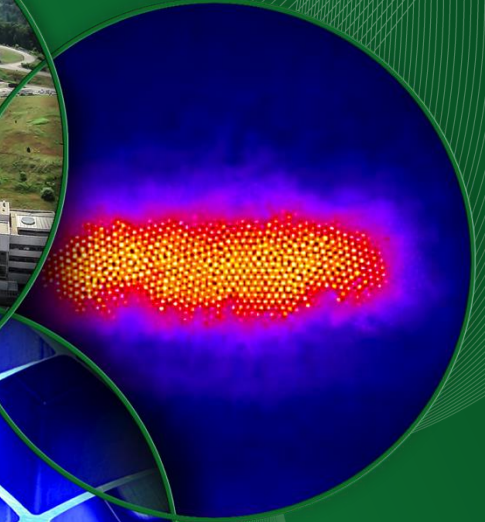


Atomic Pair Distribution Function (PDF) Analysis

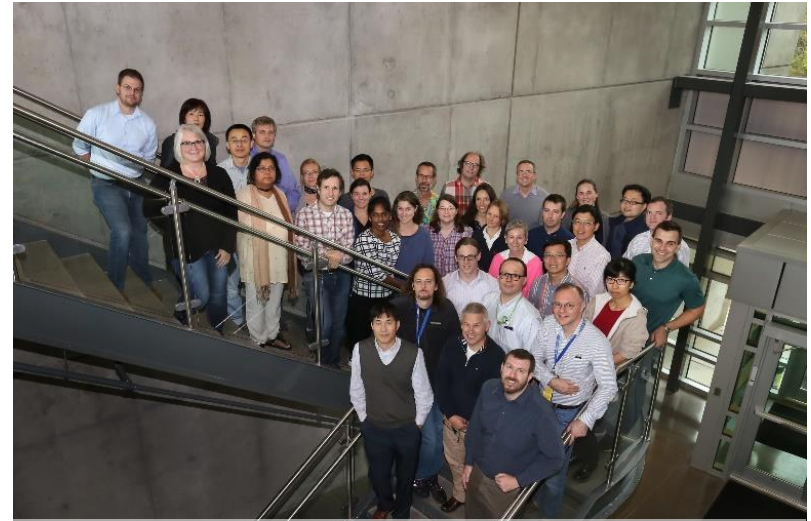
2018 Neutron and X-ray Scattering
School

Katharine Page
Diffraction Group
Neutron Scattering Division
Oak Ridge National Laboratory
pagekl@ornl.gov

ORNL is managed by UT-Battelle
for the US Department of Energy



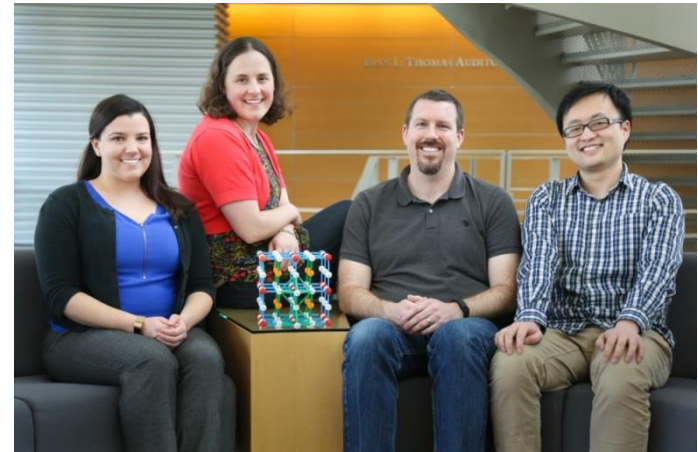
Diffraction Group, ORNL



Diffraction Group
Group Leader: Matthew Tucker

About Me (an Instrument Scientist)

- 2004: BS in Chemical Engineering University of Maine
- 2008: PhD in Materials, UC Santa Barbara
- 2009/2010: Director's Postdoctoral Fellow; 2011-2014: Scientist, Lujan Neutron Scattering Center, Los Alamos National Laboratory (LANL)
- 2014-Present: Scientist, Neutron Scattering Division, Oak Ridge National Laboratory (ORNL)



Dr. Liu, his wife (Dr. Luo), and baby boy on the way

Dr. Olds, AJ, and Max



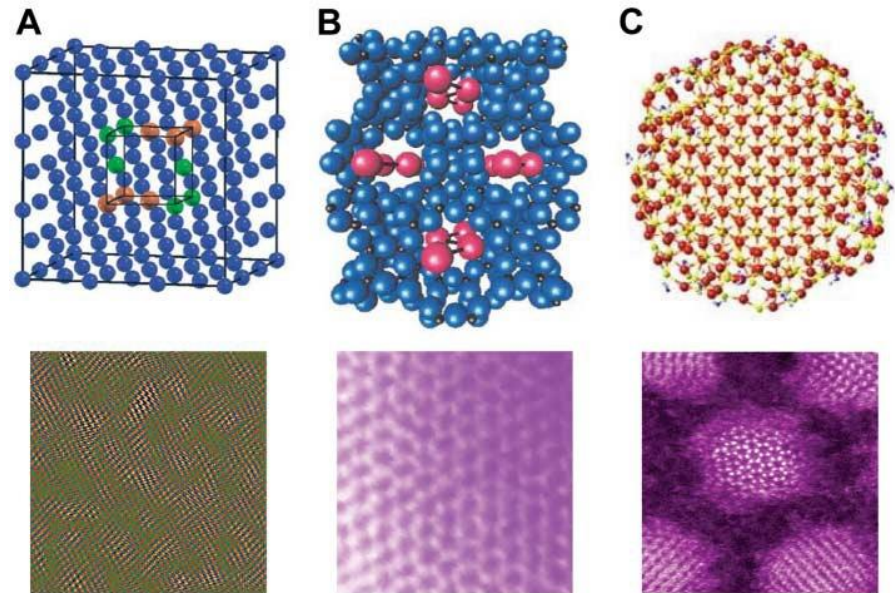
Dr. Page, Wriston, and Abbie

Dr. Usher-Ditzian and Ellie



What is a *local structure*?

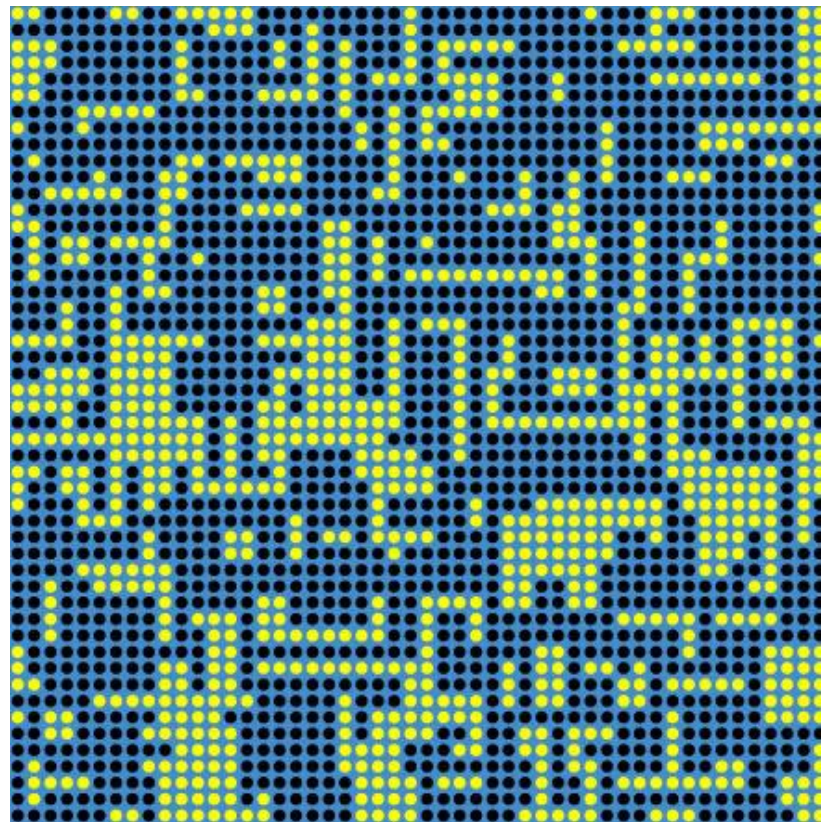
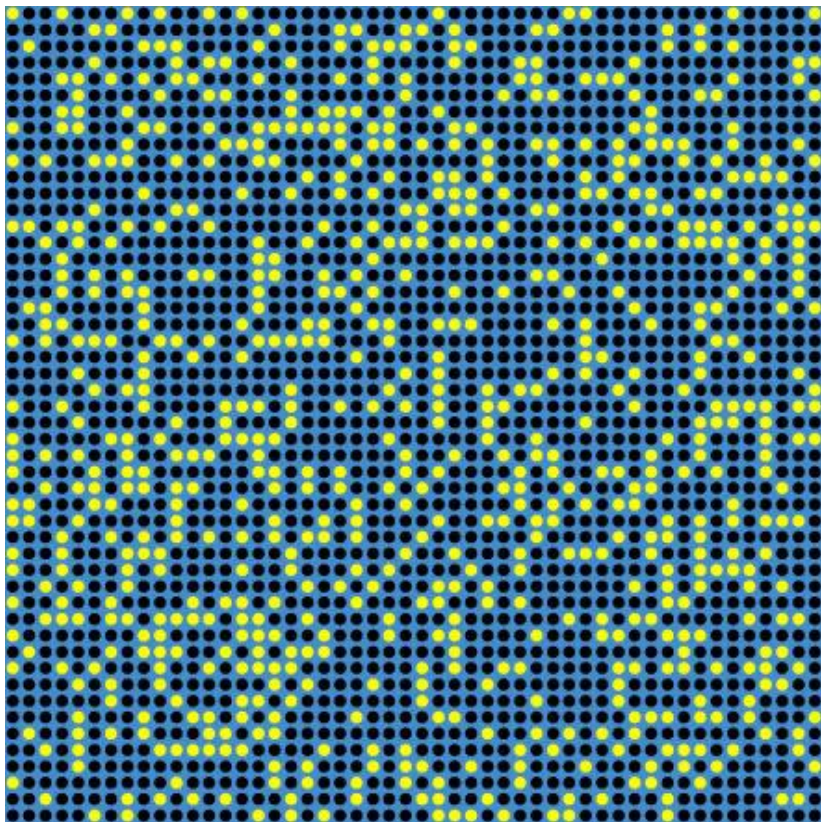
- **Disordered materials:** The interesting properties are often governed by the defects or local structure
- **Non crystalline materials:** Amorphous solids and polymers
- **Nanostructures:** Well defined local structure, but long-range order limited to few nanometers (poorly defined Bragg peaks)



S.J.L. Billinge and I. Levin, **The Problem with Determining Atomic Structure at the Nanoscale**, *Science* **316**, 561 (2007).

D. A. Keen and A. L. Goodwin, **The crystallography of correlated disorder**, *Nature* **521**, 303–309 (2015).

What is *total* scattering?



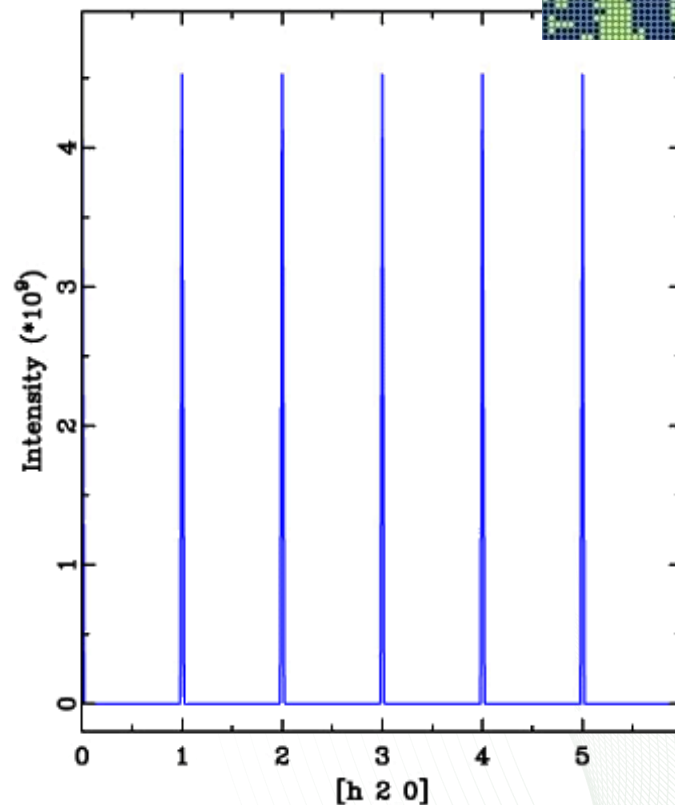
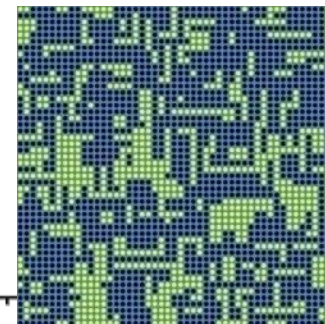
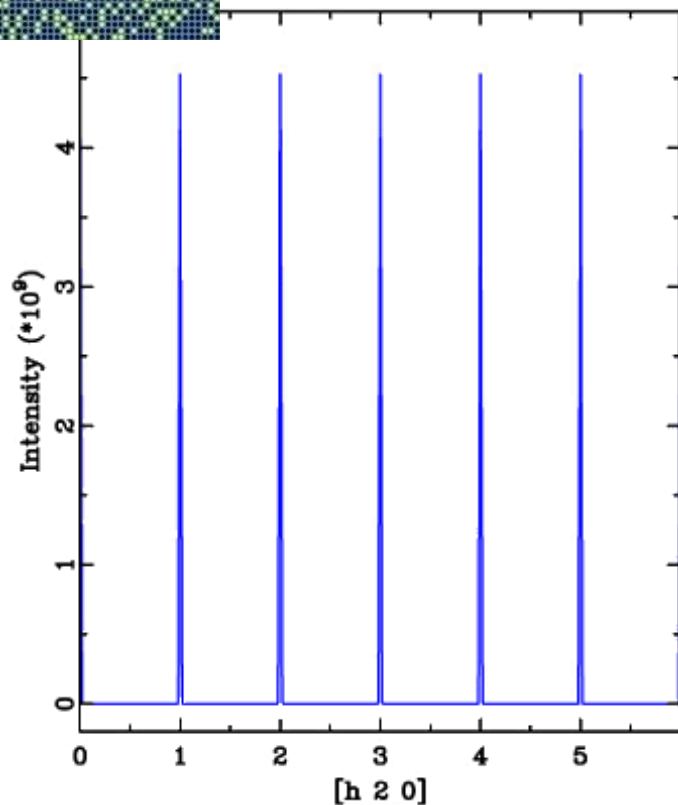
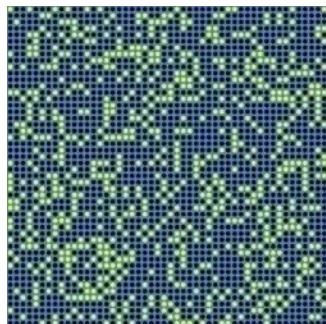
Cross section of 50x50x50 unit cell model crystal consisting of 70% blue atoms and 30% *vacancies*.

Properties might depend on vacancy ordering!

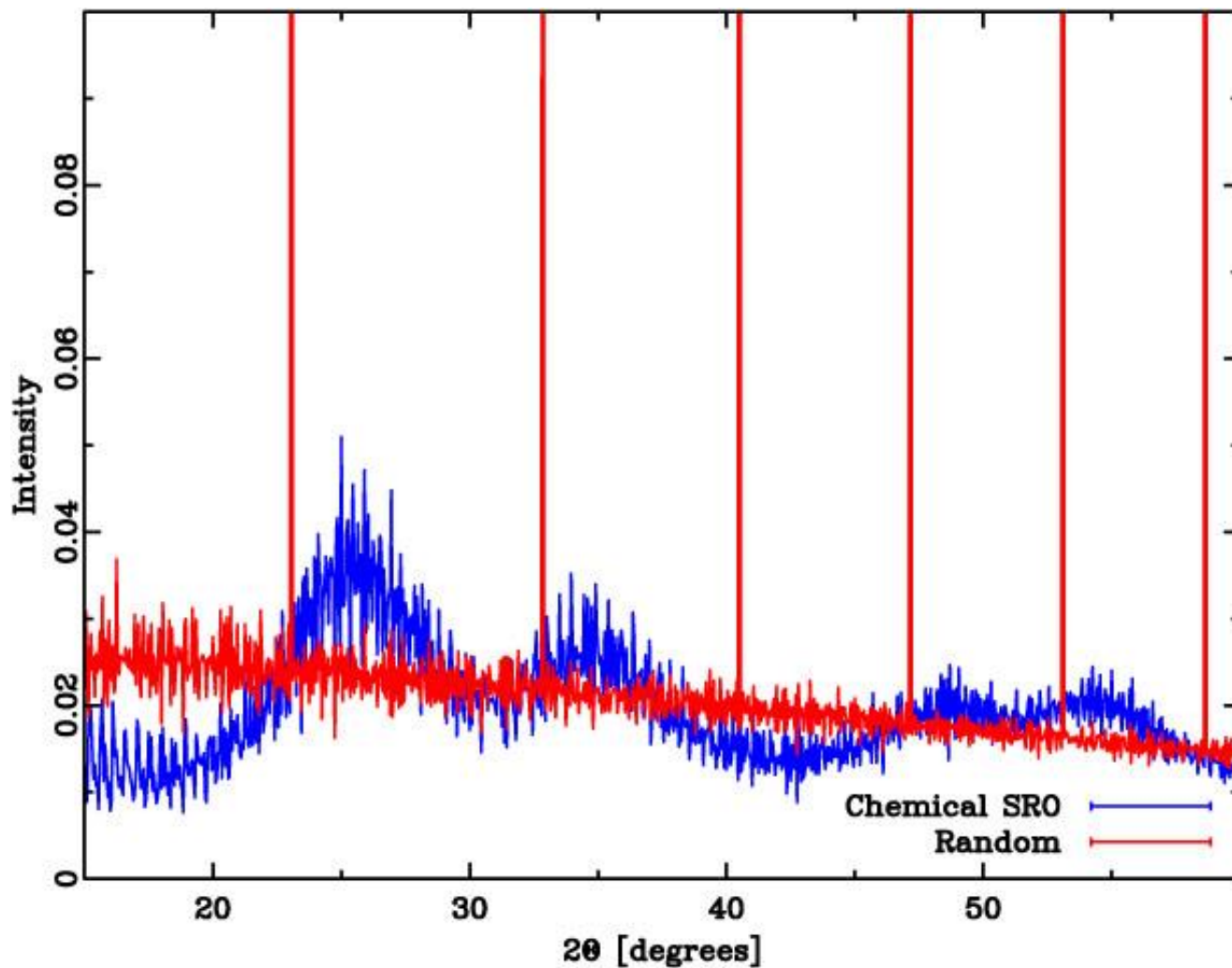
Courtesy of Thomas Proffen

Bragg peaks are “blind”

Bragg scattering: Information about the *average* structure, e.g. average positions, displacement parameters and occupancies.

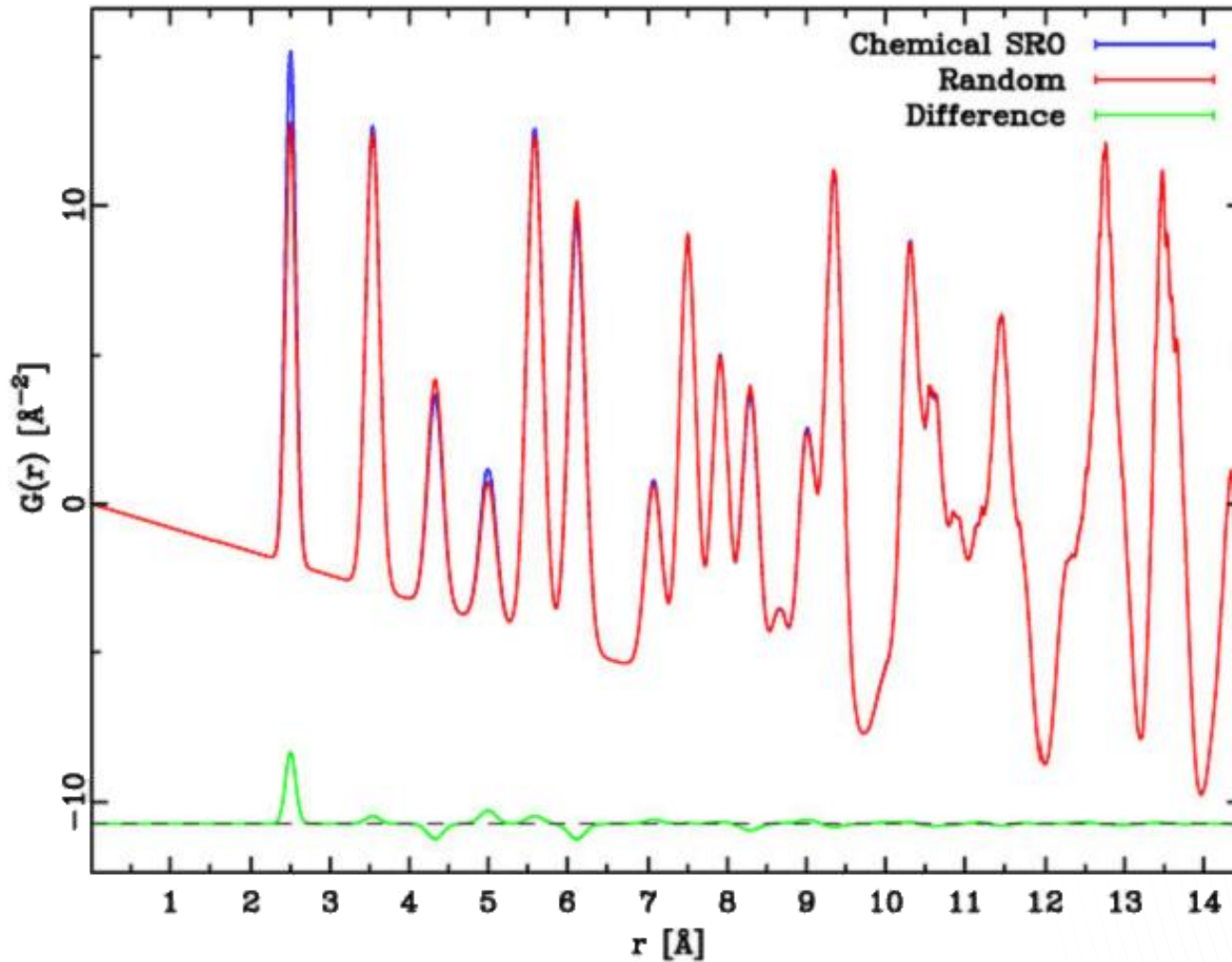


How about powder diffraction?



Courtesy of Thomas Proffen

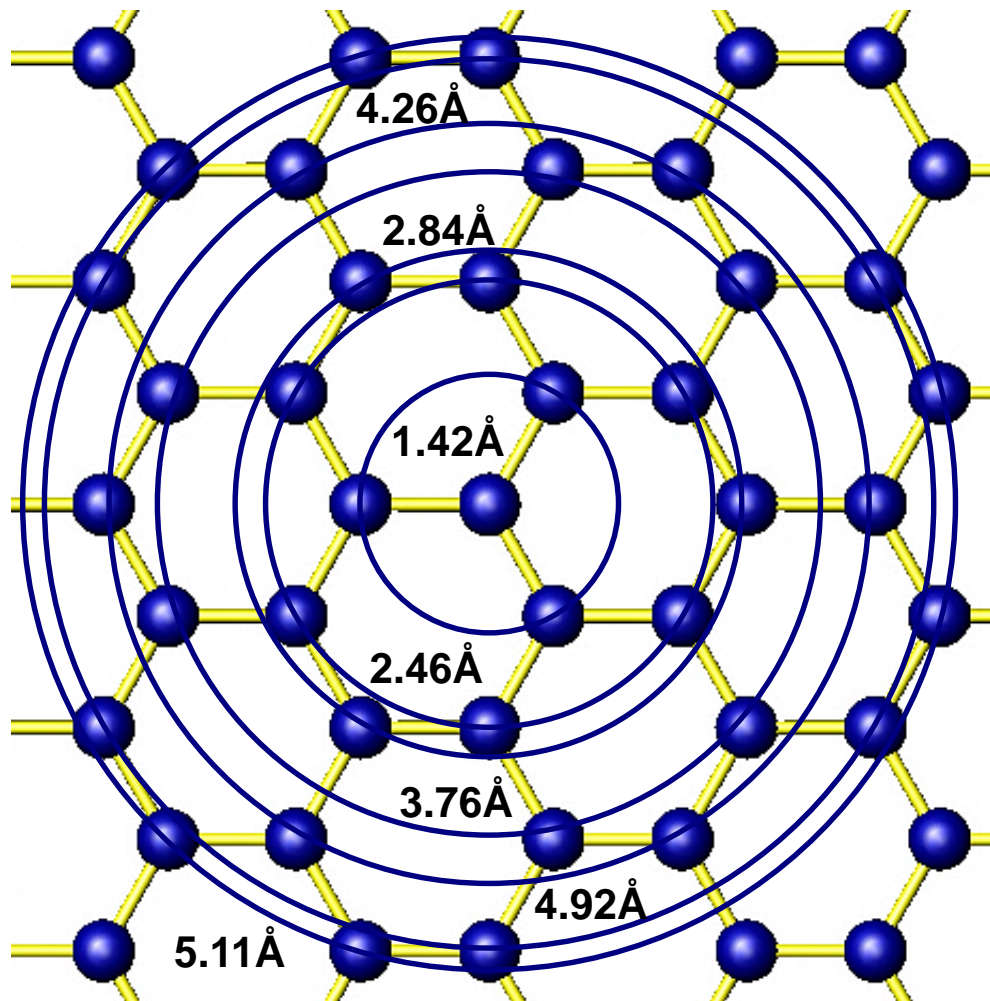
The Pair Distribution Function



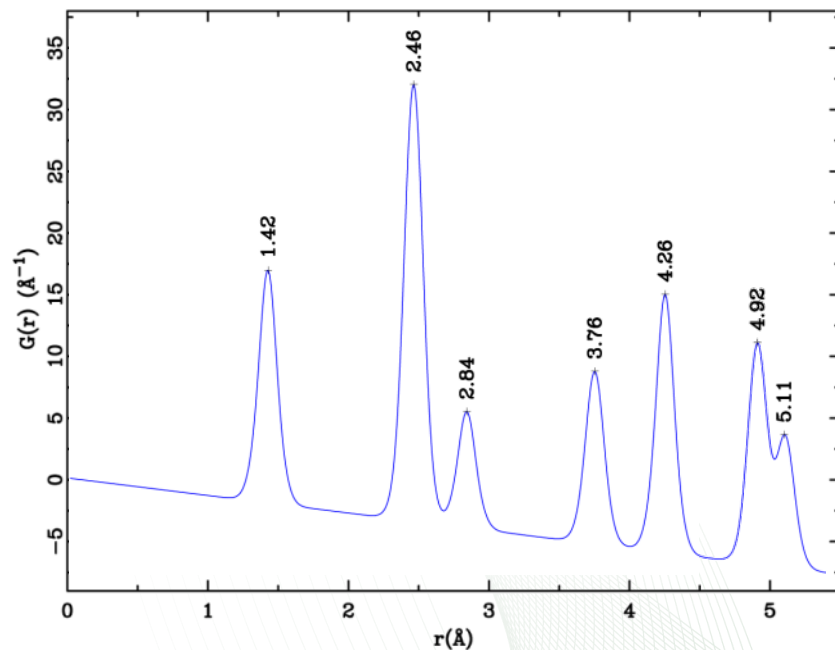
The PDF is the Sine-Fourier transform of the total scattering (Bragg and diffuse) diffraction pattern

Th. Proffen, Analysis of occupational and displacive disorder using the atomic pair distribution function: a systematic investigation, *Z. Krist.* **215**, 661 (2000).

What is a PDF?



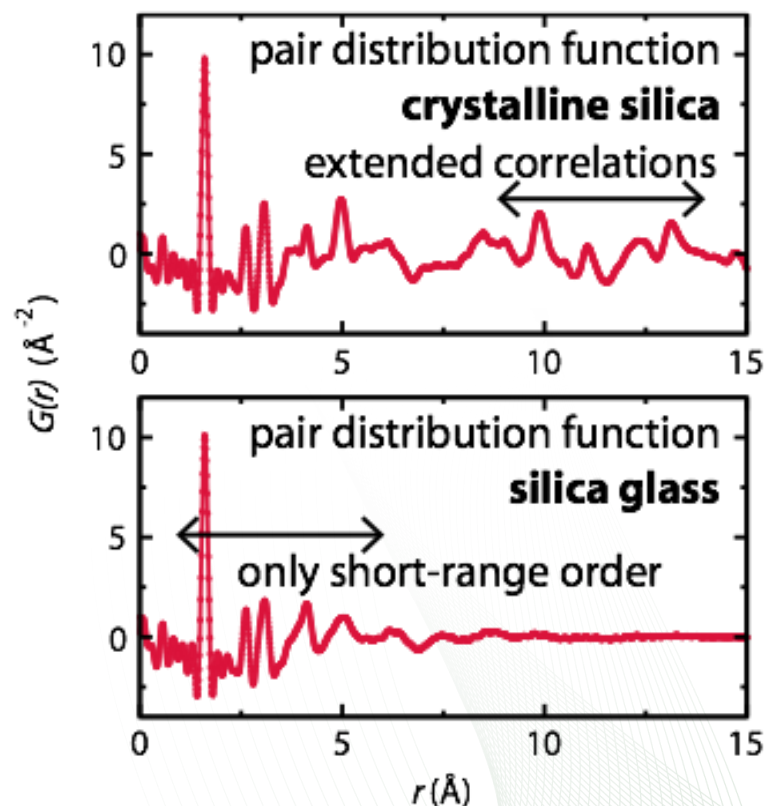
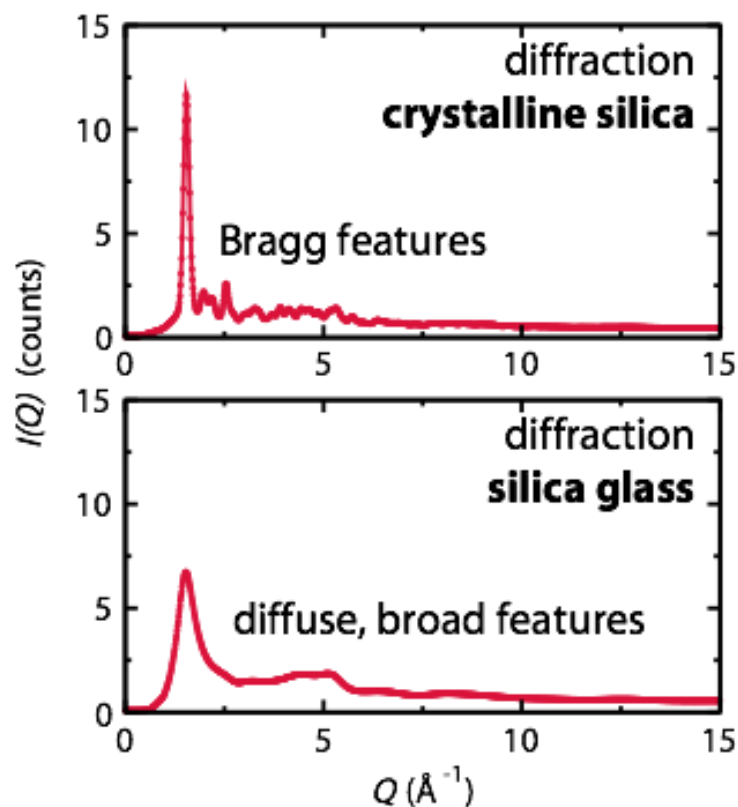
The Pair Distribution Function (PDF) gives the probability of finding an atom at a distance “ r ” from a given atom.



Pair Distribution Function

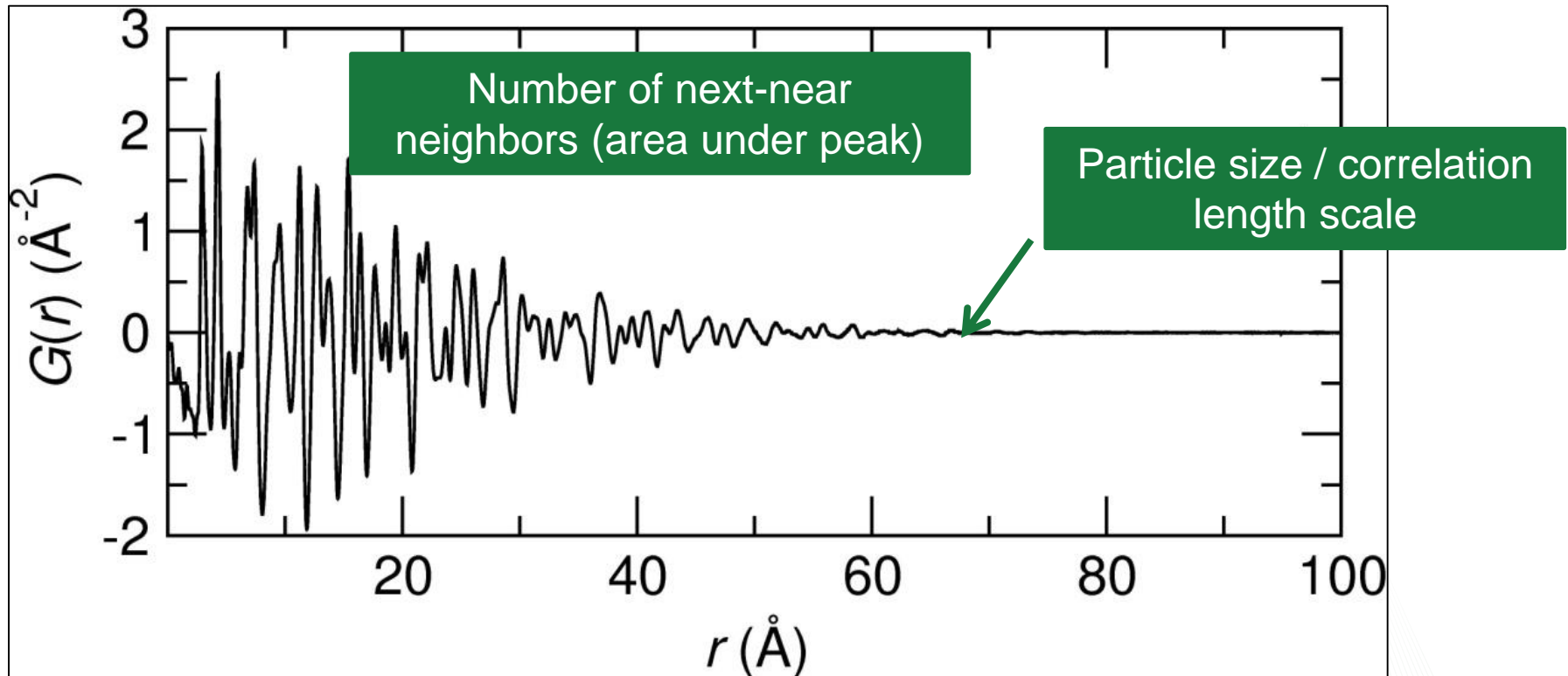
Sine-Fourier transform of **all** scattered neutron/X-ray intensity (crystalline and amorphous)

→ *Experimental, ensemble, real-space, atom-atom histogram*



Pair Distribution Function

Inter-atomic distances

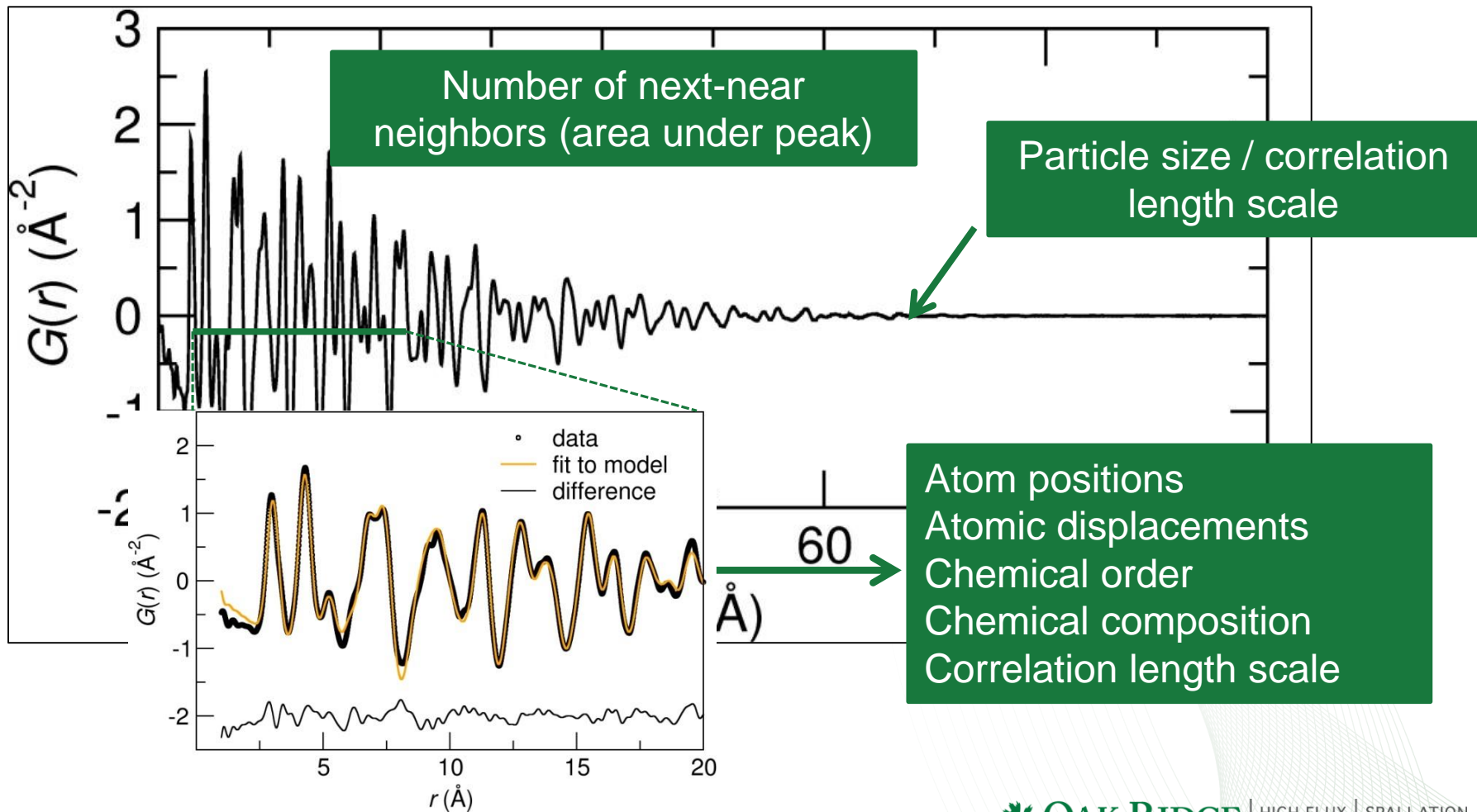


PDF analysis → Average and short-range structure for disordered crystalline materials, nanomaterials, and amorphous materials

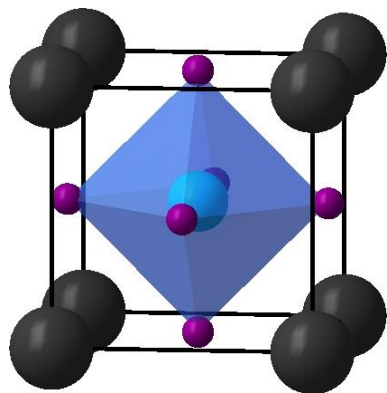
Pair Distribution Function

Quantitative analysis: fitting a model to the data over specific ranges

Inter-atomic distances

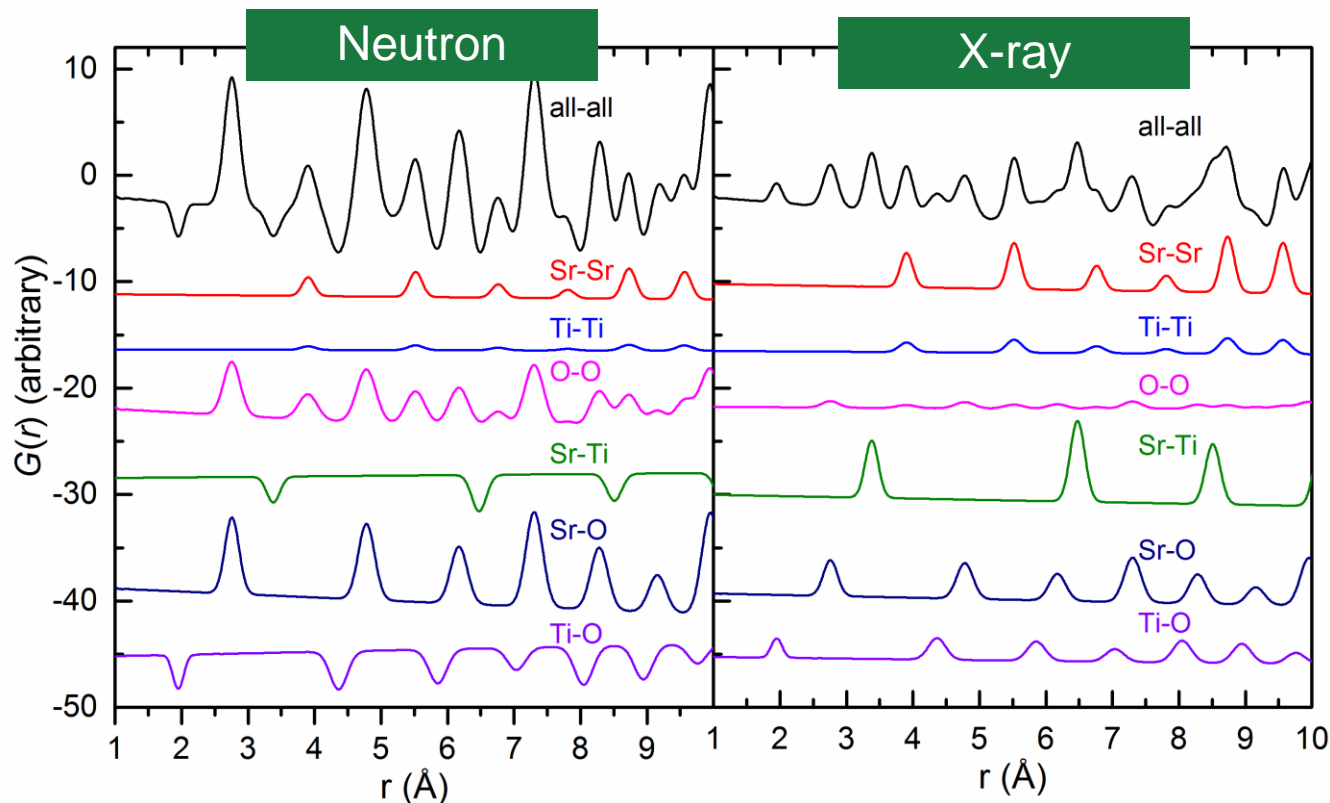


Partial PDFs



SrTiO₃

$s(s+1)/2$ partial structure factors characterize a system containing s species



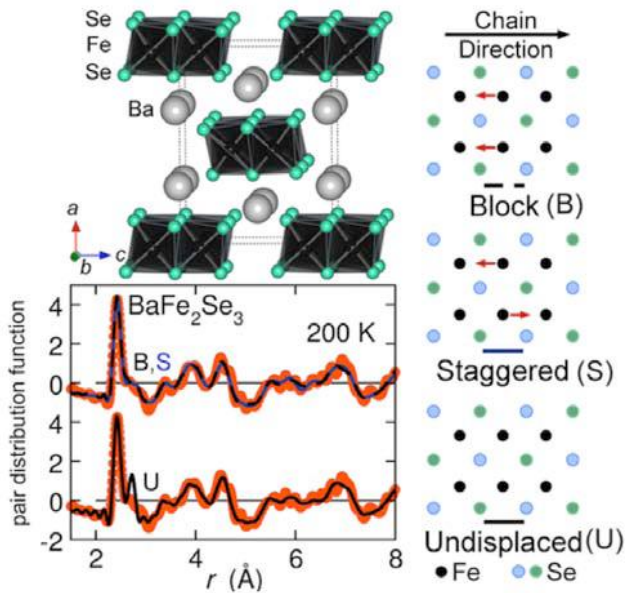
Neutron and x-ray PDFs are often highly complementary!

What types of studies can be done with the PDF technique?

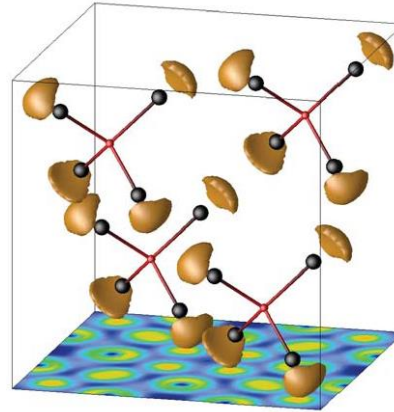


Local distortions *via* PDF

Local dipoles
Local Jahn-Teller distortions
Frustrated lattices
Orbital ordering
etc.

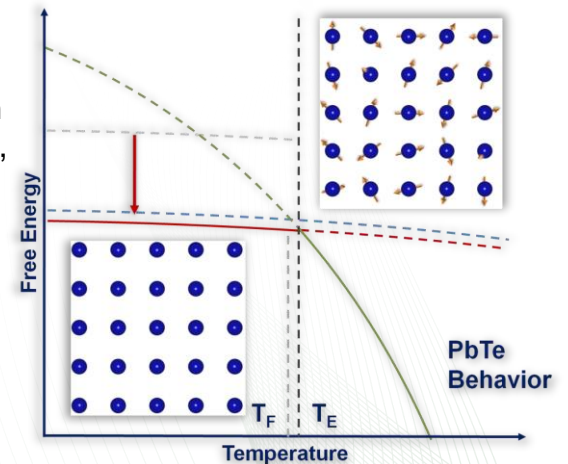


D. Louca, *et al.*, Suppression of superconductivity in Fe pnictides by annealing; a reverse effect to pressure, *Phys. Rev. B* **84**, 054522 (2011).



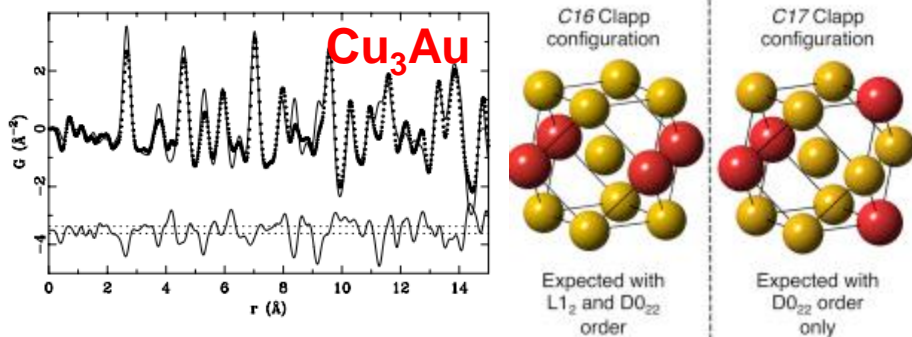
D. P. Shoemaker, *et al.*, Reverse Monte Carlo neutron scattering study of the 'ordered-ice' oxide pyrochlore $Pb_2Ru_2O_{6.5}$, *J. Phys.: Condens. Matter* **23** (2011).

E. Bozin, *et al.*, Entropically Stabilized Local Dipole Formation in Lead Chalcogenides, *Science* **330**, 1660 (2010).

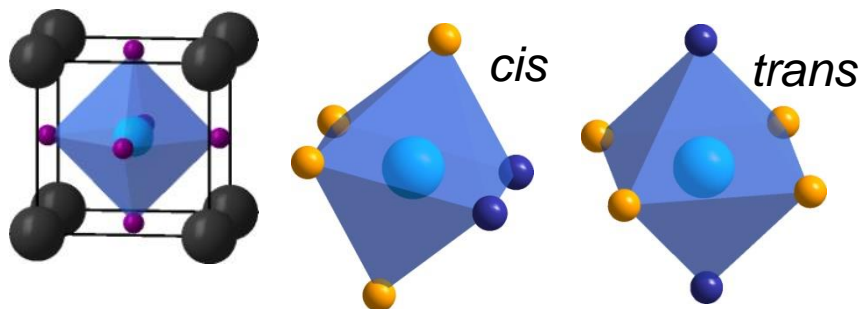


Chemical Short Range order *via* PDF

Th. Proffen, V. Petkov, S. J. L. Billinge, and T. Vogt,
Chemical short range order obtained from the atomic pair
distribution function, *Z. Kristallogr.* **217**, (2002) 47–50.



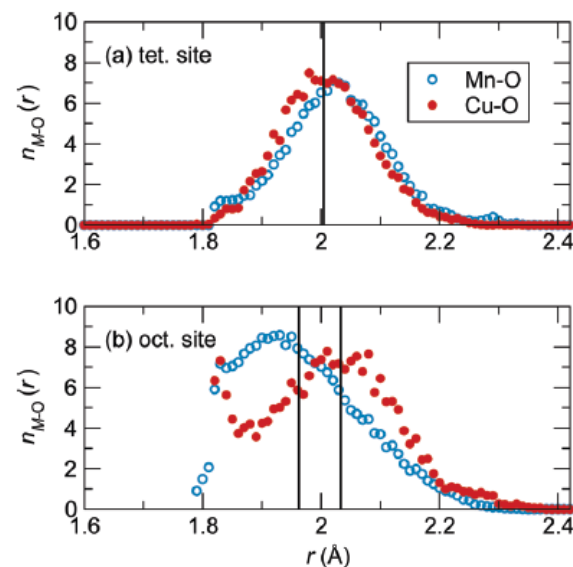
L.R. Owen, H.Y. Playford, H.J. Stone and M.G. Tucker,
Analysis of short-range order in Cu_3Au using X-ray pair
distribution functions. *Acta Materialia* (2017) 125, 15-26.



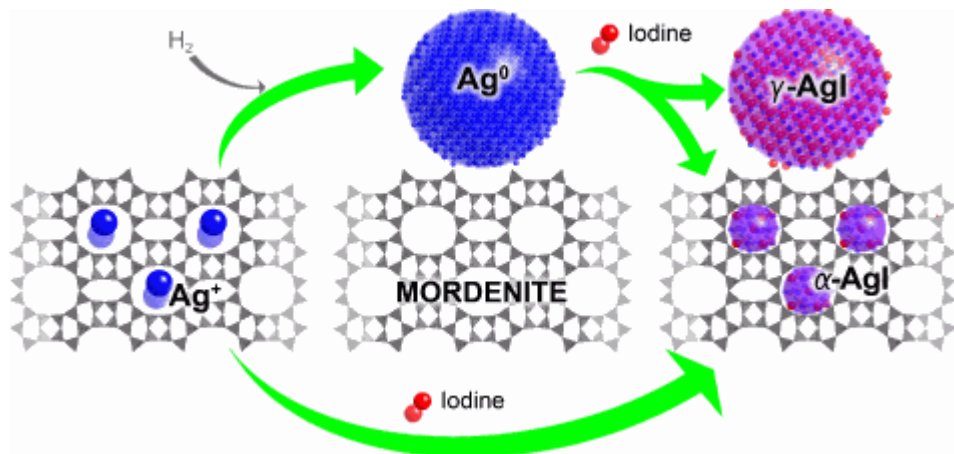
K. Page, *et al.*, Local atomic ordering in BaTaO_2N studied by
neutron pair distribution function analysis and density functional
theory, *Chem. Mater.* **19** (2007) 4037-4042.

Substitution effects
Chemical clustering
Ion-specific local environments
Vacancy ordering

D. P. Shoemaker, J. Li, and R. Seshadri,
Unraveling Atomic Positions in an Oxide Spinel with
Two Jahn-Teller Ions: Local Structure Investigation
of CuMn_2O_4 , *J. Am. Chem. Soc.* **131**, 11450 (2009).

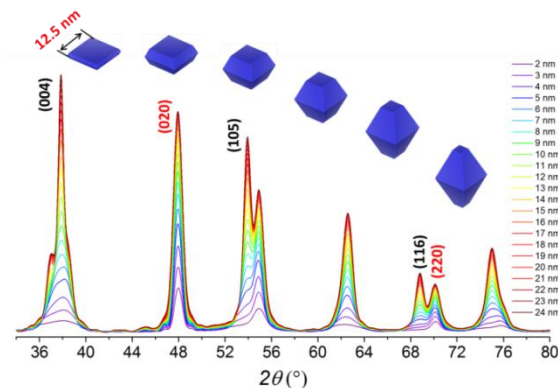


Nanomaterial structure *via* PDF

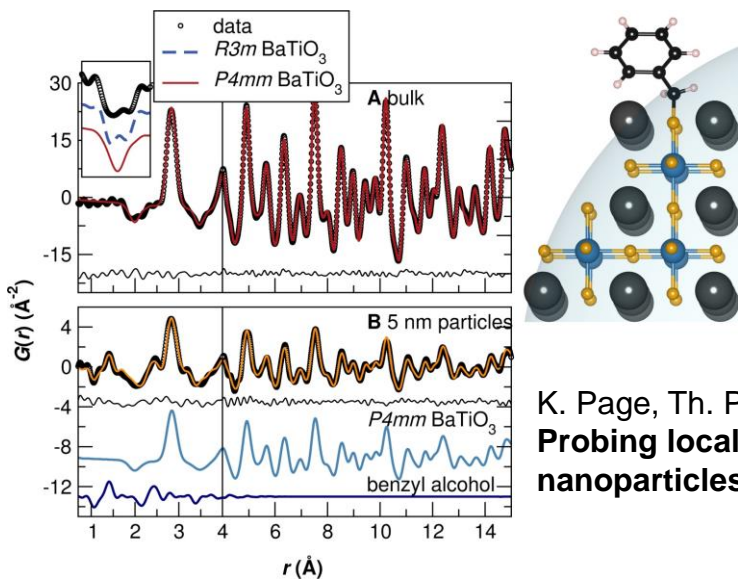
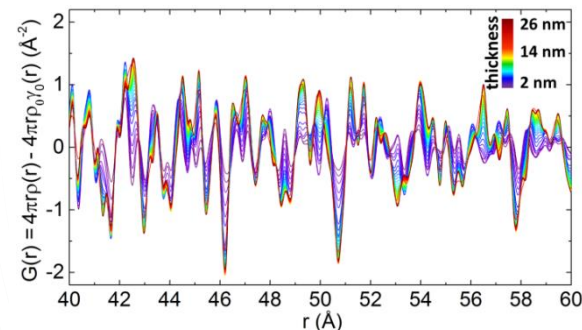


K. W. Chapman, P. J. Chupas, and T. M. Nenoff, **Radioactive Iodine Capture in Silver-Containing Mordenites through Nanoscale Silver Iodide Formation**, *J. Am. Chem. Soc.* 132, 8897 (2010).

Finite size/shape effects
Surface/Interface structure
Nanostructure polymorphs
Growth and transformation



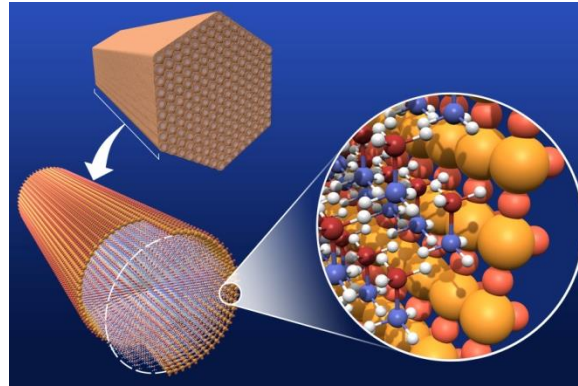
J. Liu et al, **Quantitative analysis of the morphology of {101} and {001} faceted anatase TiO₂ nanocrystals**, *Chem. Mater.* 29, 5591–5604 (2017).



K. Page, Th. Proffen, M. Niederberger, and R. Seshadri, **Probing local dipoles and ligand structure in BaTiO₃ nanoparticles**, *Chem. Mater.* 22 (2010), 43864391.

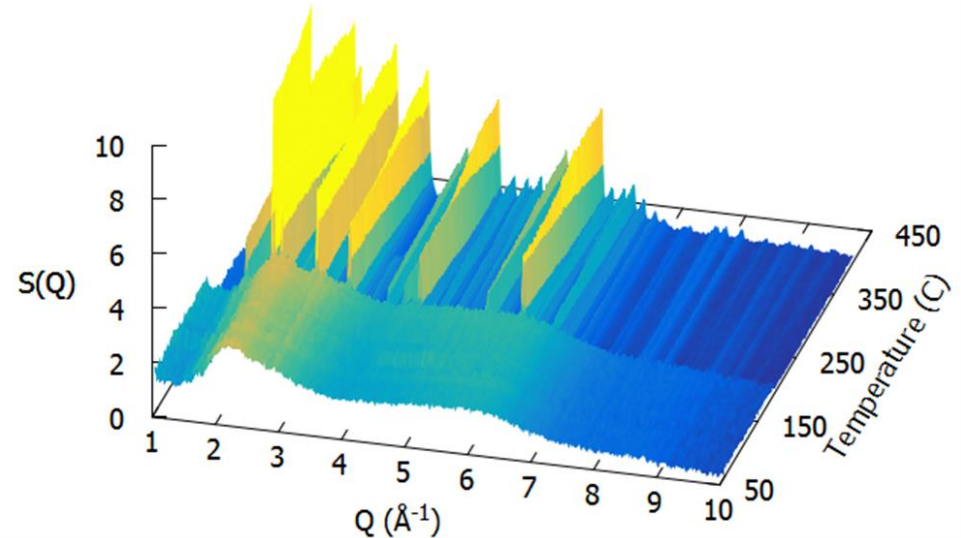
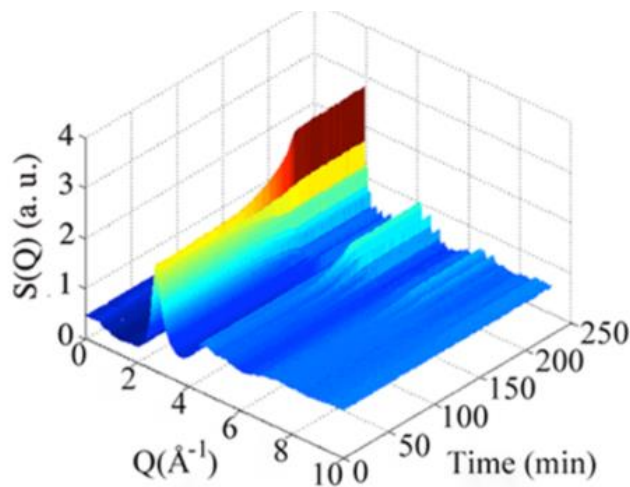
Amorphous structures *via* PDF

Glasses
Liquids
Concretes
Adsorbed gases
etc.



Kim, H.; Proffen, T.; Chupas, P. J.; Karkamkar, A.; Hess, N. J.; Autrey, T., **Determination of structure and phase transition of light element nanocomposites in mesoporous silica: case study of NH_3BH_3 in MCM-41**, *J. Am. Chem. Soc.* **2009**, *131*(38) 13749-13755.

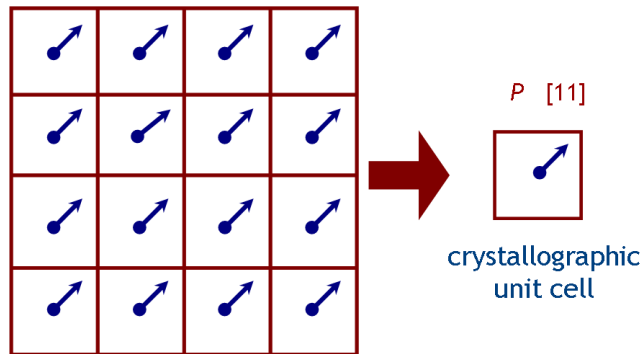
S. Lan, X.. Wei, J. Zhou, Z. Lu, X. Wu, M. Feygenson, J. Neufeind, X. Wang, **In situ study of crystallization kinetics in ternary bulk metallic glass alloys with different glass forming abilities**, *Applied Physics Letters* **105** (2014) 201906.



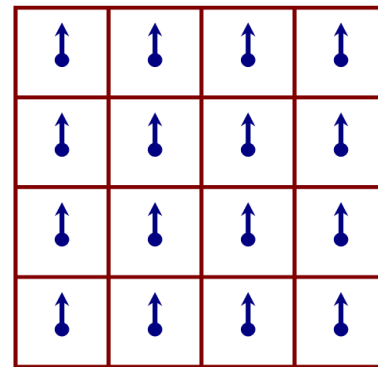
H.-W. Wang; L. L. Daemen, M. C. Cheshire, M. K. Kidder, A. G. Stack, L. F. Allard, J. Neufeind, D. Olds, J. Liu, and K. Page, **Synthesis and structure of synthetically pure and deuterated amorphous (basic) calcium carbonates**, *Chem. Commun.*, 53, 2942-2945 (2017).

Example: Local structure in BaTiO₃

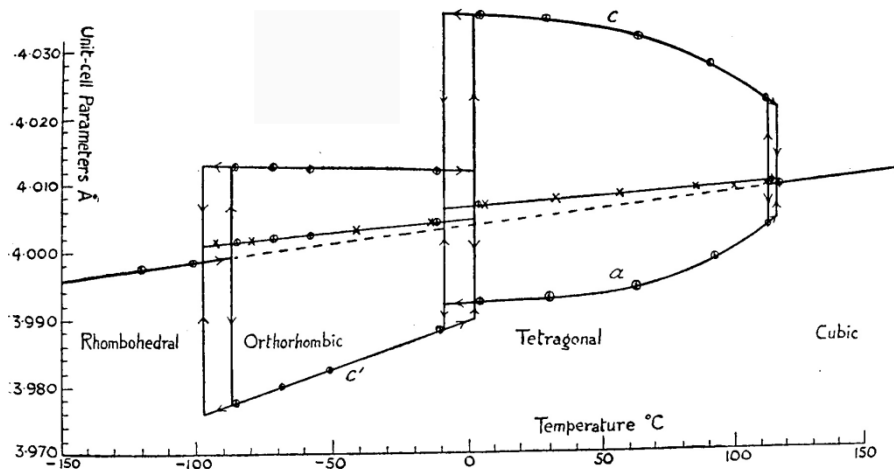
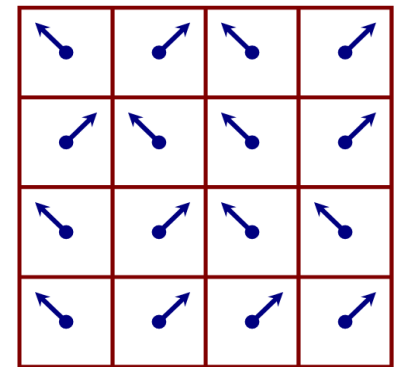
BaTiO₃: Ferroelectric oxide, a rhombohedral (*R3m*) ground state and a room temperature tetragonal (*P4mm*) structure



displacive model



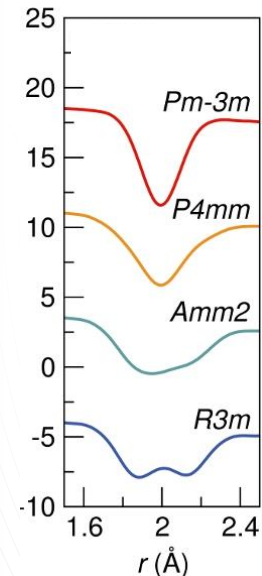
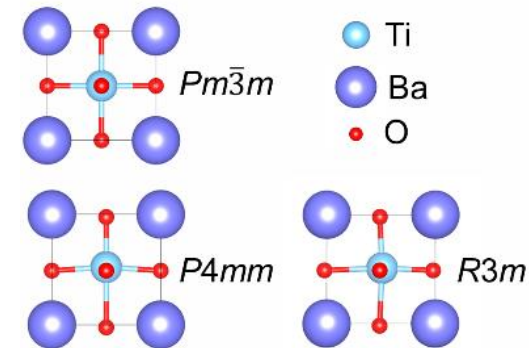
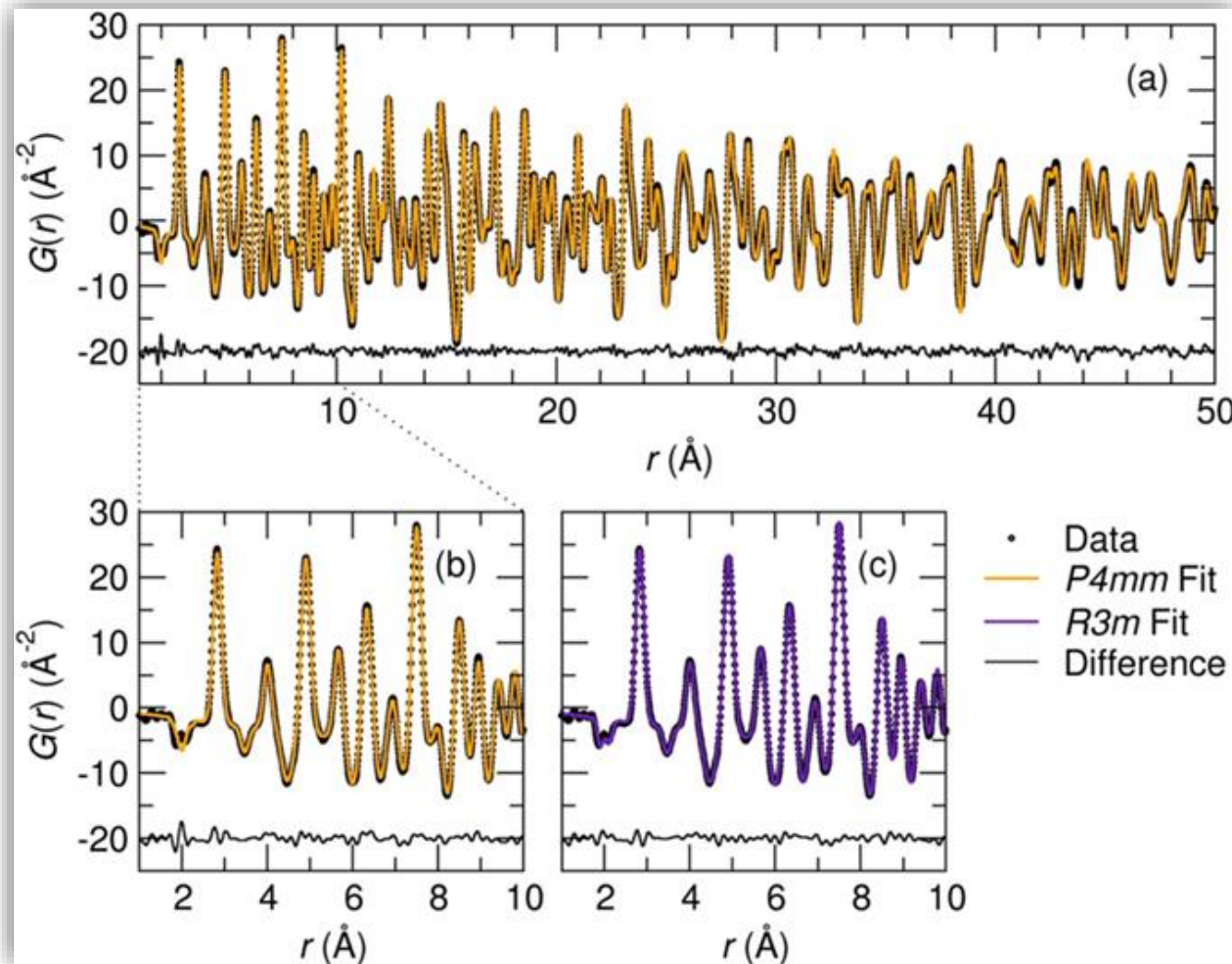
order-disorder model



Displays order-disorder phenomena:
room temperature local structure
known to have rhombohedral pair-
pair correlations

Example: Local structure in BaTiO₃

BaTiO₃: At room temperature, locally has a split (*R3m*) first Ti-O peak, displaying classic order-disorder behavior



K. Page, T. Kolodiazhnyi, T. Proffen, A. K. Cheetham, and R. Seshadri, *Phys. Rev. Lett.* **101**, 205502 (2008).

Usher *et al.*, *J. Appl. Phys.* **120**, 184102 (2016)
 Senn *et al.*, *Phys. Rev. Lett.* **116**, 207602 (2016)

Example: High Voltage Spinel Cathode

$\text{LiNi}_{0.5}\text{Mn}_{1.5}\text{O}_4$

High operating voltage (~ 4.7 V versus Li^+/Li) and facile three dimensional lithium ionic conductivity Zhong *et al.*, 1997; Ohzuku *et al.*, 1999

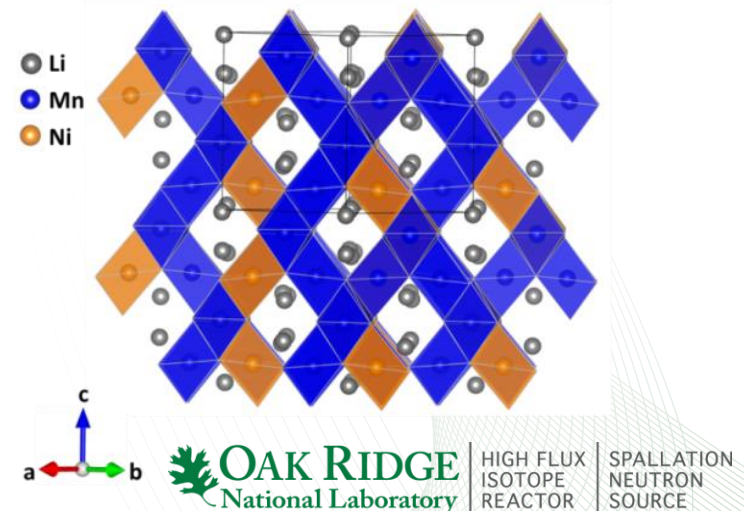
Two distinct polymorphs are known: Ni/Mn cation ordering strongly impacts electrochemical performance Idemoto *et al.*, 2003; Zhong *et al.*, 1997

(1) Disordered phase (S.G. $Fd-3m$), where Ni/Mn are randomly distributed at the 16d site via high temperature solid state reaction

(2) Long-range cation ordered phase (S.G. $P4_332$ or $P4_132$) via extended post-annealing at 700 °C to 600 °C

Nature of the cation ordering and detailed mechanisms are still debated

Kunduraci & Amatucci, 2006; Kunduraci *et al.*, 2006; Kim *et al.*, 2004; Ma *et al.*, 2010; Moorhead-Rosenberg *et al.*, 2015



Example: High Voltage Spinel Cathode

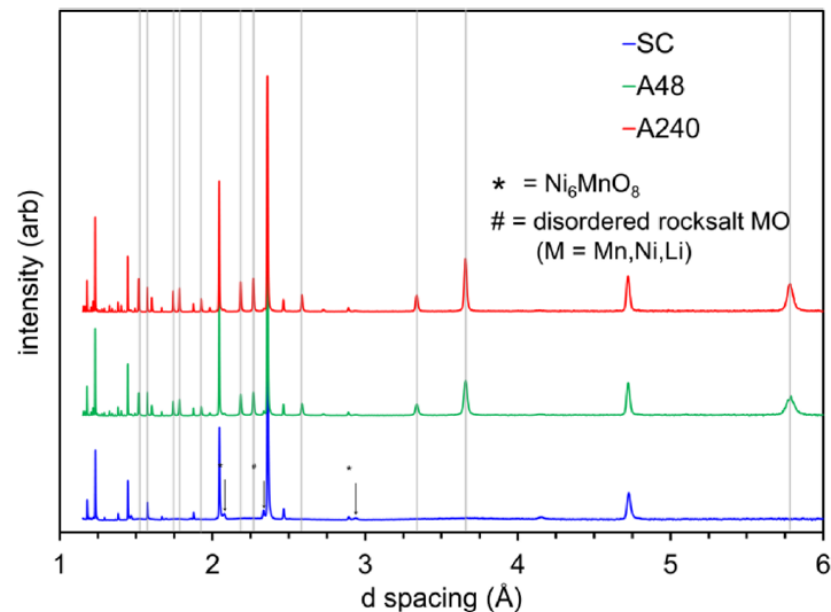
Cation ordering examined at the POWGEN Beamline, SNS

Slow Cooled (SC): 8 hours at 900°C, 1.5°C/min cooling

Fast Cooled (FC): 8 hours at 900°C, 5°C/min cooling

Annealed (A48): 48 hours at 700°C

Annealed (A240): 240 hours at 700°C



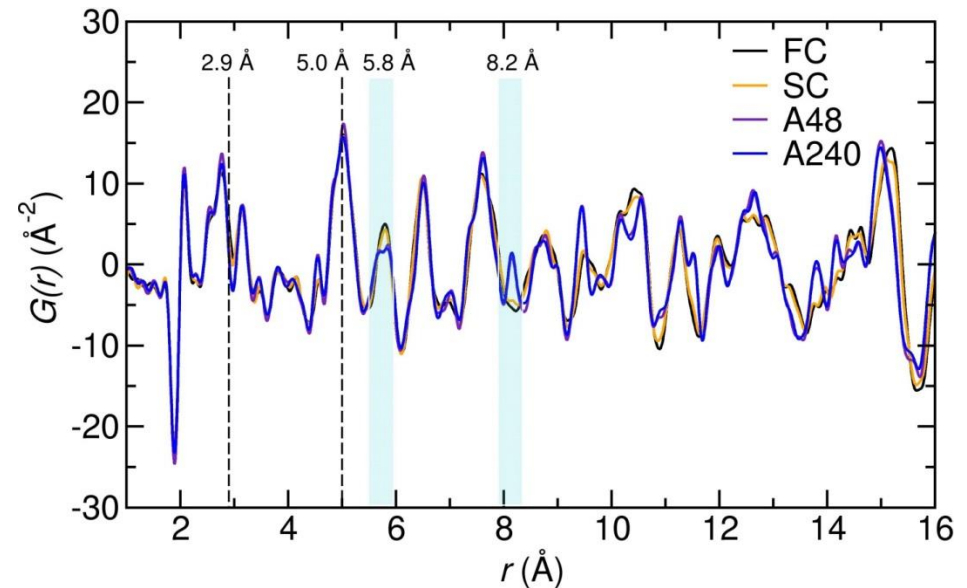
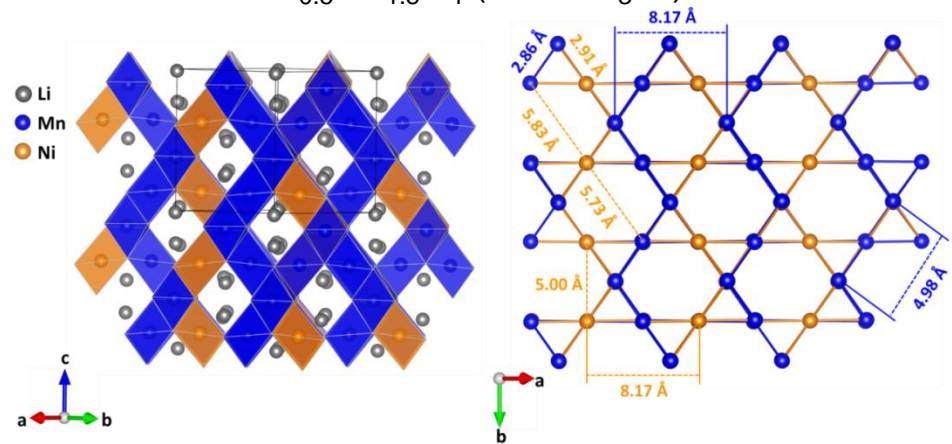
Ni/Mn ordering: large nuclear scattering length contrast between nickel ($b = 10.3$ fm) and manganese ($b = -3.73$ fm).

Moorhead-Rosenberg, Z., Huq, A., Goodenough, J. B. & Manthiram, A. **Electronic and Electrochemical Properties of $\text{Li}_{1-x}\text{Mn}_{1.5}\text{Ni}_{0.5}\text{O}_4$ Spinel Cathodes As a Function of Lithium Content and Cation Ordering**, *Chem. Mater.* 27, 6934-6945, 2015.

We can tell a lot by *looking* at PDF data

- Local atomic structures almost identical up to 5 Å (two nearest B-site neighbors)
- Sample structures diverge after that
- Annealed samples: two distinguishable sets of Ni/Mn pairs at third nearest Ni/Mn neighbor distance
- By fourth nearest Ni/Mn neighbor samples are very distinct

ordered $\text{LiNi}_{0.5}\text{Mn}_{1.5}\text{O}_4$ (S.G. $P4_332$)

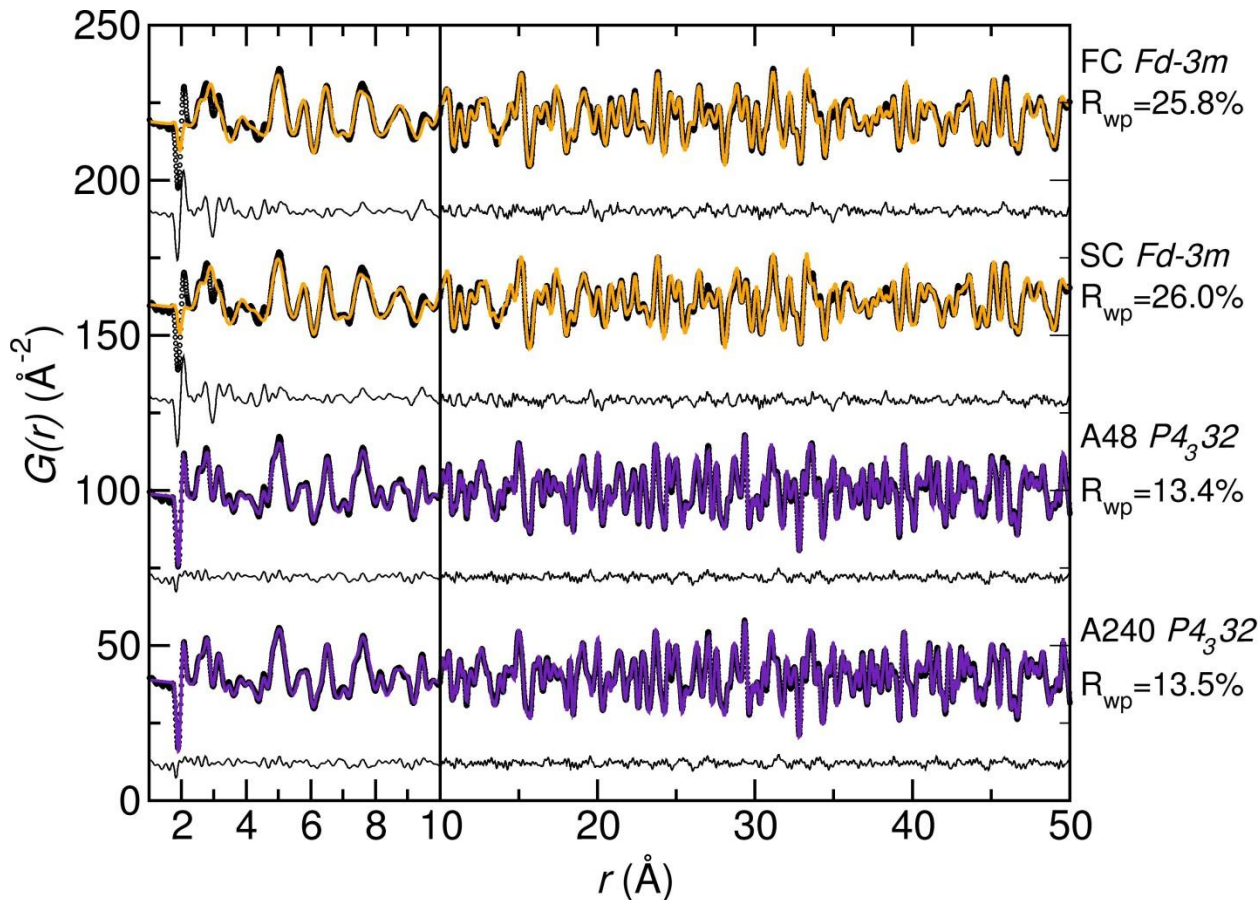


J. Liu, A. Huq, Z. Moorhead-Rosenberg, A. Manthiram, and K. Page, **Nanoscale Ni/Mn ordering in the high voltage spinel cathode $\text{LiNi}_{0.5}\text{Mn}_{1.5}\text{O}_4$** , *Chemistry of Materials*, 28, 19, 6817–6821, 2016.

POWGEN: 3 hours per frame, Q-range: 1 – 30 \AA^{-1}

Example: High Voltage Spinel Cathode

Modeling the local structure with average structure models determined from Rietveld refinement

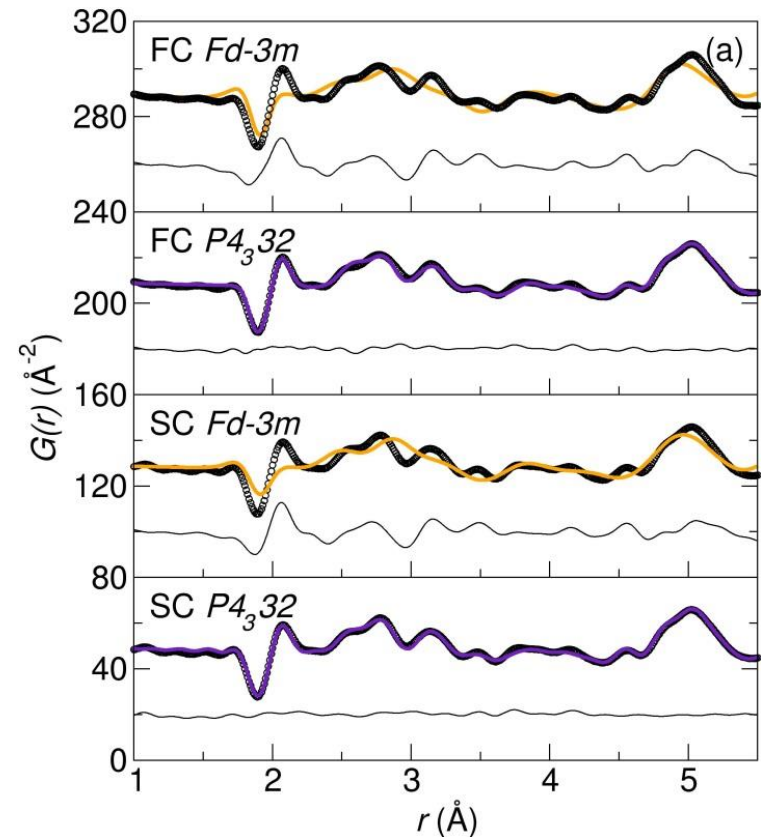


- First 10 \AA of non-annealed sample data (FC and SC) differ *strongly* from long-range crystallographic structures
- Local atomic structure of annealed samples (A48 and A240) are well-described by the long-range crystallographic structures

Example: High Voltage Spinel Cathode

Modeling the local structure with alternate models

- Least squares refinements of the ordered structure model ($P4_332$) were carried out in a manner that $4b$ site (Ni site) and $12d$ site (Mn site) occupancies are allowed to refine simultaneously but with site multiplicity constraints
- Over 1 to 5 Å range the ordered models ($P4_332$) with Mn/Ni site mixing provide much better fits for local PDF profiles
- **Ni/Mn are locally well-ordered in the long-range “disordered” samples**
- **But up to what length scale?**



J. Liu, A. Huq, Z. Moorhead-Rosenberg, A. Manthiram, and K. Page, **Nanoscale Ni/Mn ordering in the high voltage spinel cathode $\text{LiNi}_{0.5}\text{Mn}_{1.5}\text{O}_4$** , *Chemistry of Materials*, 28, 19, 6817–6821, 2016.

Example: High Voltage Spinel Cathode

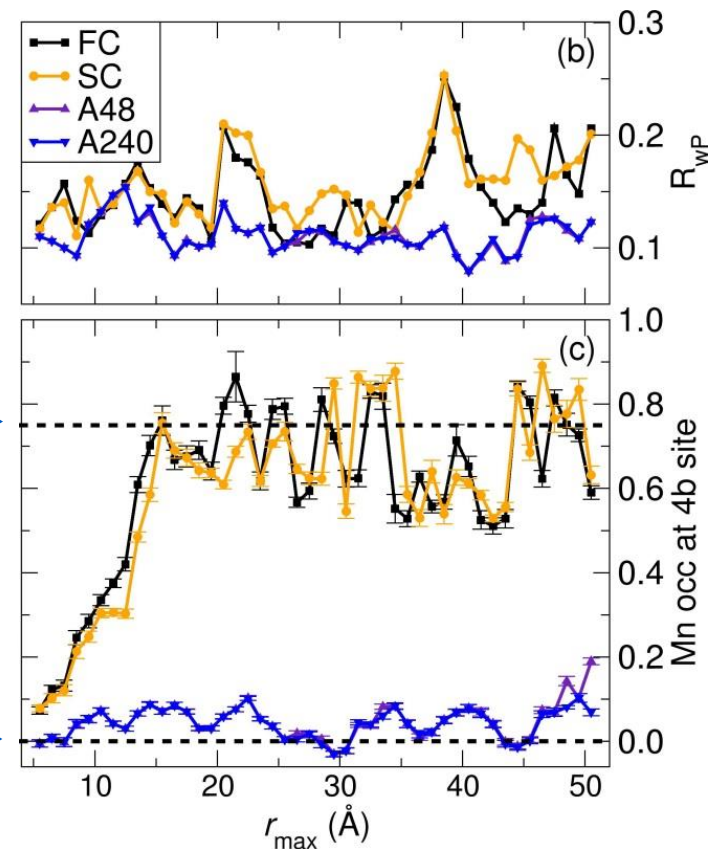
Fit the PDFs within a 4.5 Å “box” in 1 Å steps (a “box-car” refinement)

- 5% site mixing in the A48 and A240 patterns throughout the entire range
- FC and SC samples are nearly fully disordered at pair distances beyond 15.5 Å

Spinel cathode materials are distinguished by their unique correlation length scales for chemical short range ordering

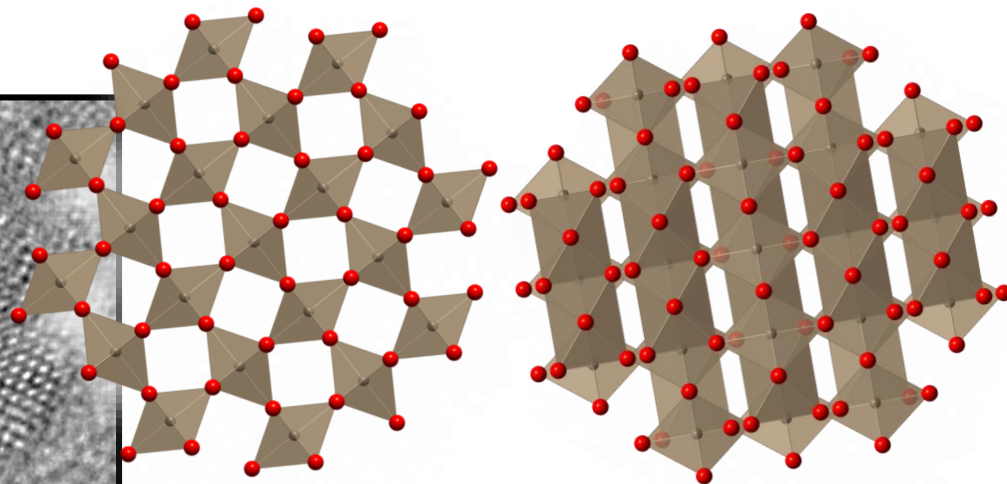
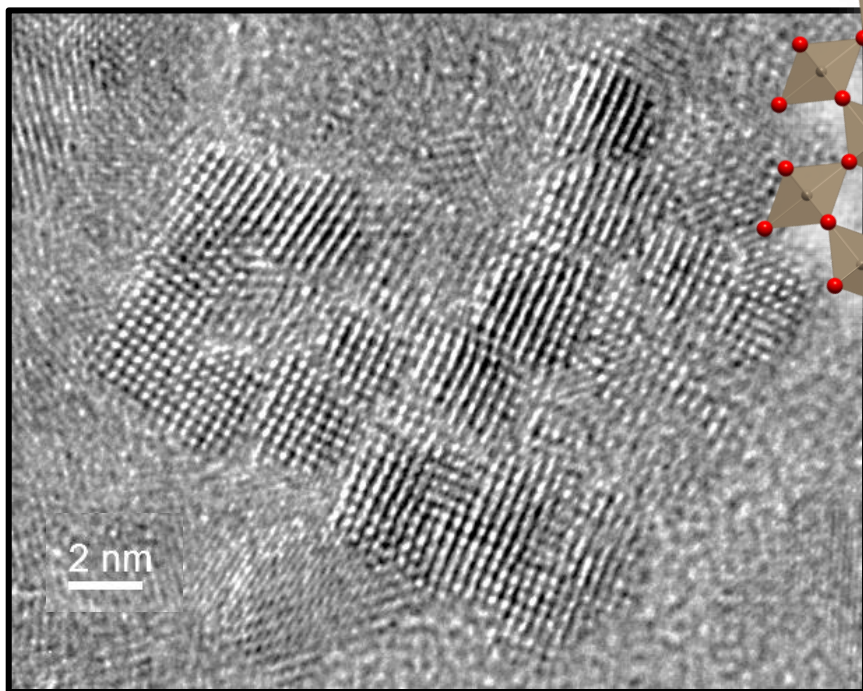
fully disordered →

fully ordered →



Example: SnO₂ Nanocrystals

~2 nm SnO₂ (cassiterite) nanocrystals capped with H₂O/OH or D₂O/OD groups



TGA suggests 2 steps dehydration.

How many layers of water are at the surface?

How is water bonded to surfaces?

What are the dynamics of dehydration?

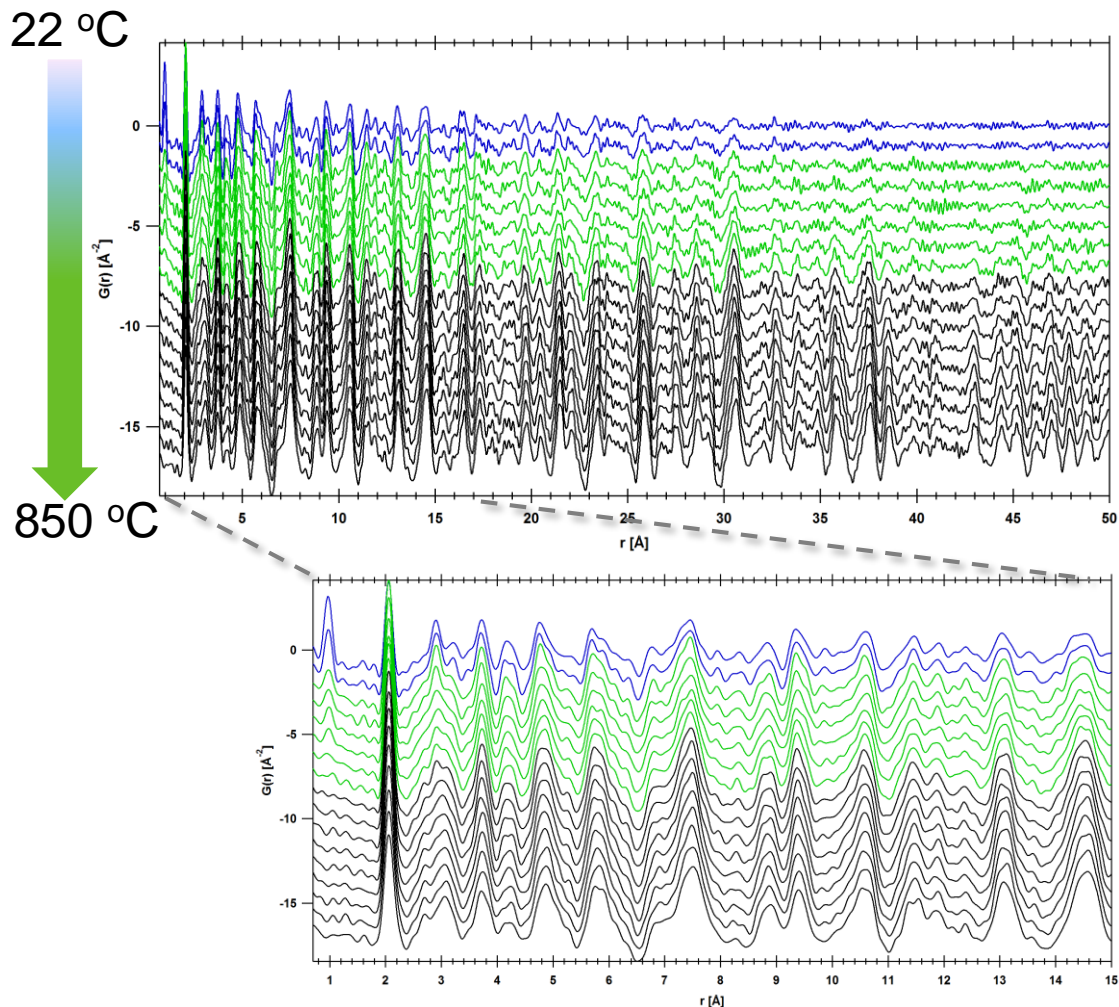
H.-W. Wang, D. J. Wesolowski, T. Proffen, L. Vıcek, W. Wang, L. F. Allard, A. I. Kolesnikov, M. Feygenson, L. M. Anovitz, and R. L. Paul, **Structure and stability of SnO₂ nanocrystals and surface-bound water species**, *J. Am. Chem. Soc.*, 135, 6885-6895, 2013.

Example: SnO₂ Nanocrystals

In situ Dehydration at NOMAD, SNS

Starting with full coverage

- Blue lines:
22 and 50 °C - L₁ + L₂ + L₃
- Green lines:
50 to 350 °C (with 50 °C increments) - L₁ + L₂
- Black lines:
400 to 850 °C (with 50 °C increments) – SnO₂ grain growth

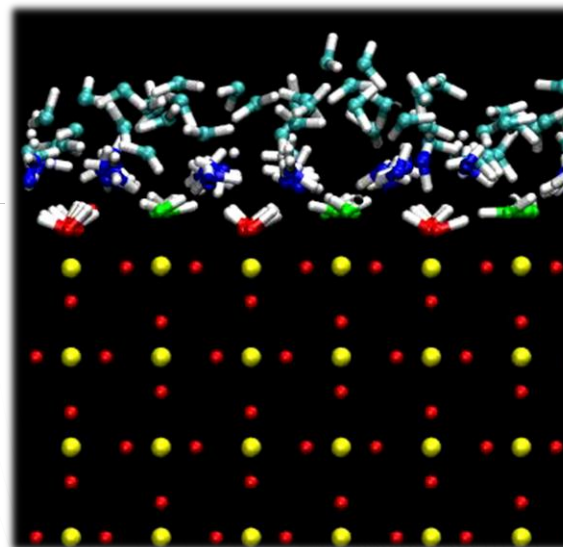
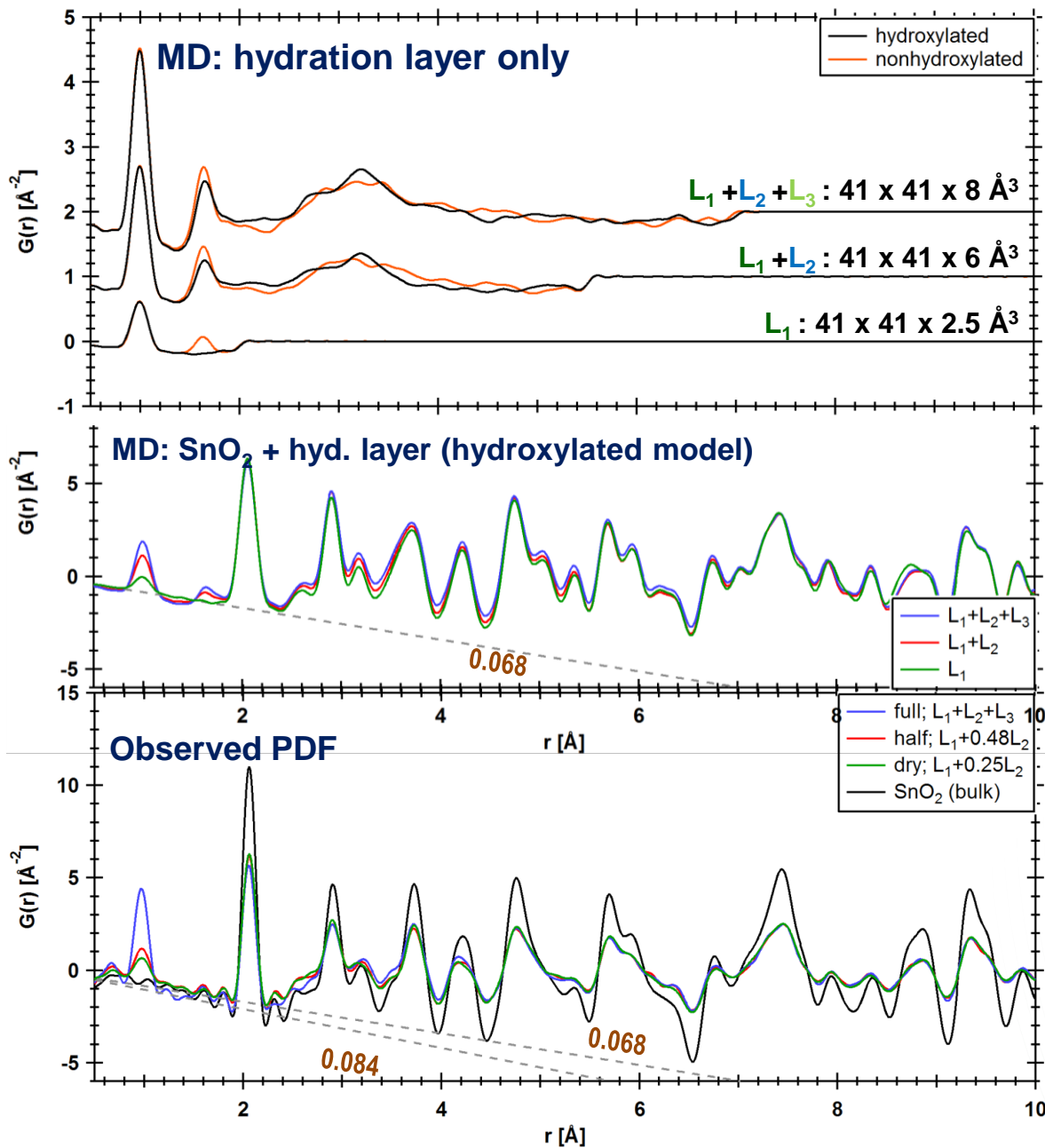


H.W. Wang, et al, *J. Am. Chem. Soc.*, 2013, 135, 6885–6895.

MD and PDF

Data is compared to Molecular Dynamics Simulation PDFs for nonhydroxylated and hydroxylated models:

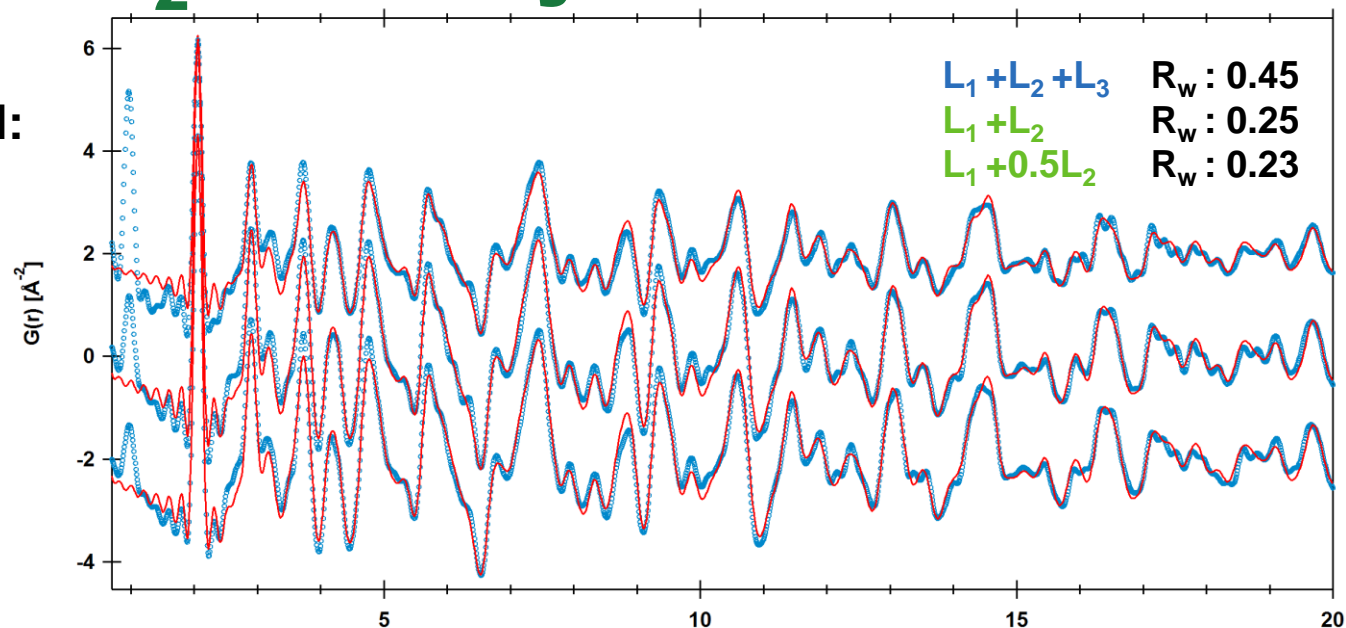
Box size: $41 \times 41 \times 23 \text{ \AA}^3$; 2592 atoms; # density = 0.068 \AA^{-3} ; $U_{\text{iso}} = 0.003 \text{ \AA}^2$



Example: SnO₂ Nanocrystals

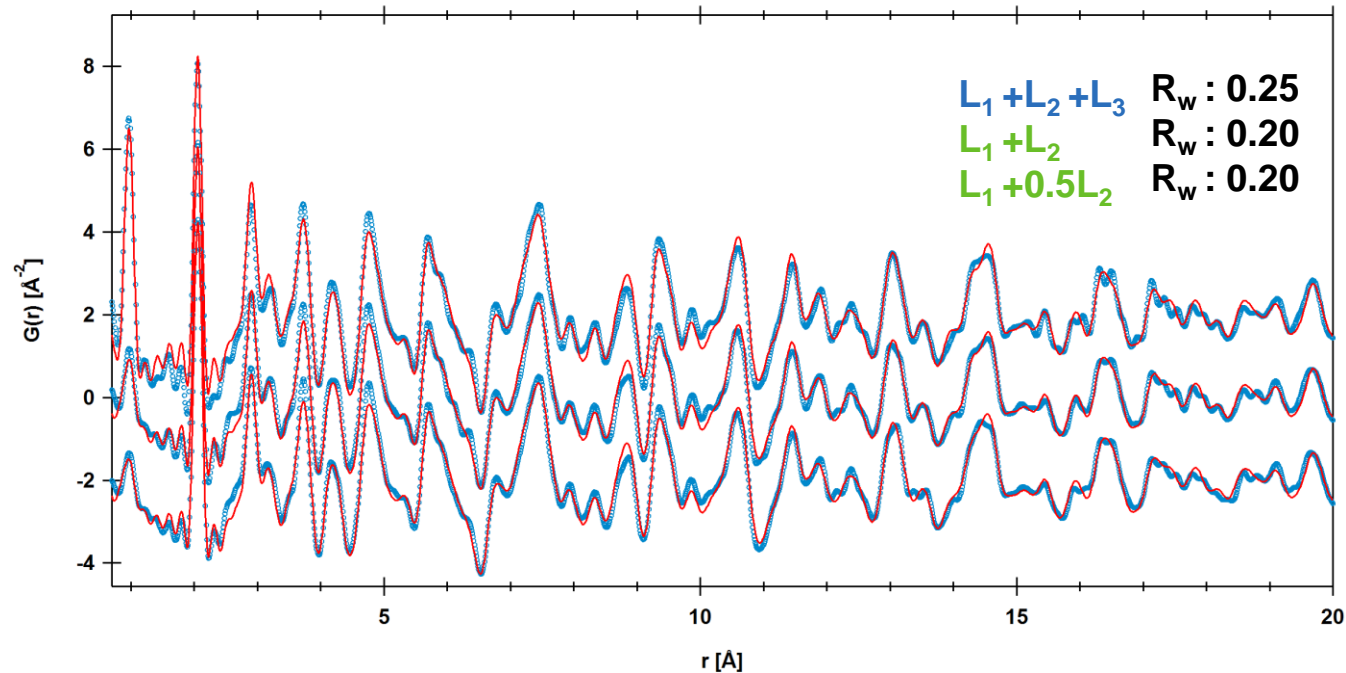
Single phase model:

SnO₂ bulk structure,
refined particle
size = ~47 Å

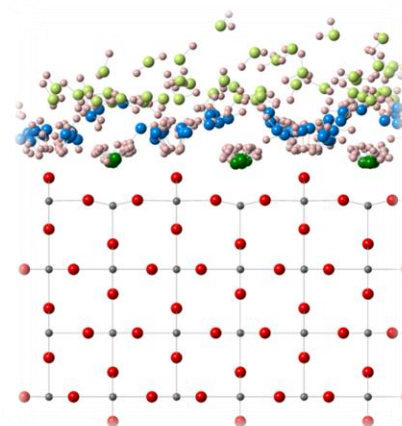
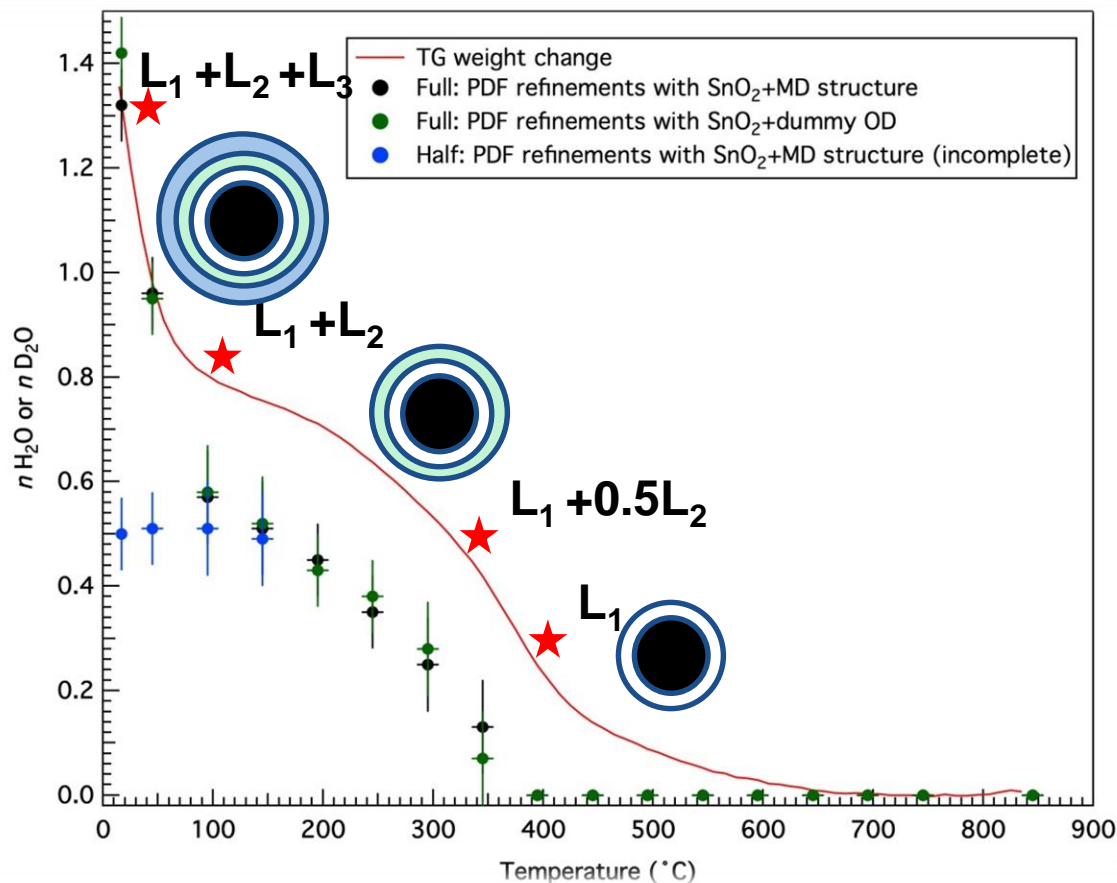


Two phase model:

SnO₂ bulk +
layered MD water
structure



Example: SnO₂ Nanocrystals



H.-W. Wang, D. J. Wesolowski, T. Proffen, L. Vlcek, W. Wang, L. F. Allard, A. I. Kolesnikov, M. Feygenson, L. M. Anovitz, and R. L. Paul, **Structure and stability of SnO₂ nanocrystals and surface-bound water species**, *J. Am. Chem. Soc.*, 135, 6885-6895, 2013.

A Few Experimental Considerations



Total Scattering Structure Function

Structure function, determined from the scattering intensity/differential cross section:

coherent scattering intensity (corrected) *scattering length (neutrons) or atomic form factor (x-rays)*

$$S(Q) = \frac{I_{coh}(Q) - \sum c_i |b_i|^2}{|\sum c_i b_i|^2} + 1 \qquad Q = \frac{4\pi \sin \theta}{\lambda}$$

Corrected for: Container & background scattering, self-absorption, etc.

Normalized by: Incident flux, number of atoms, square of the scattering length/form factor

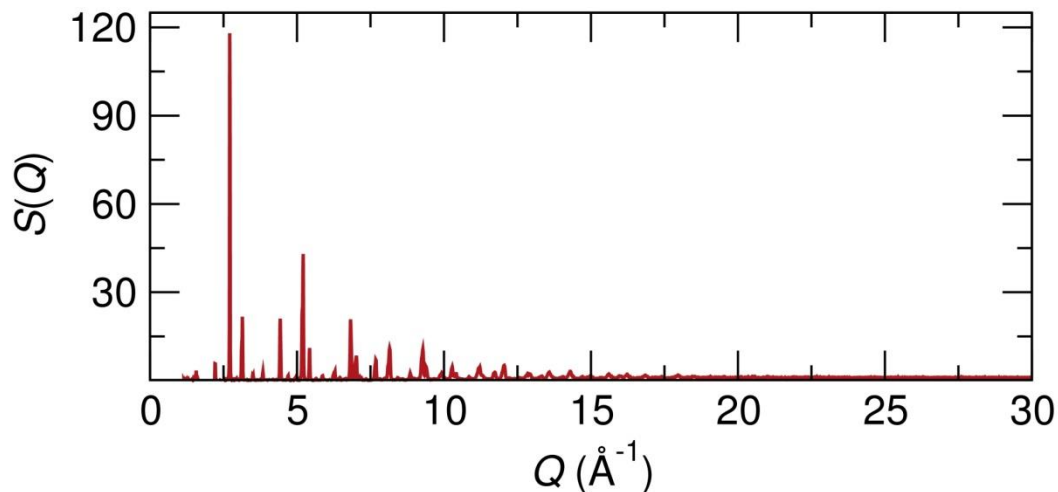
For unambiguous derivation of this derivation and relationship to other forms:

C. Farrow and S. J. L. Billinge, *Acta Cryst.* (2009) A65, 232–239.

D. A. Keen, *J. Appl. Cryst.* 34 (2001) 172-177.

The Experimental PDF

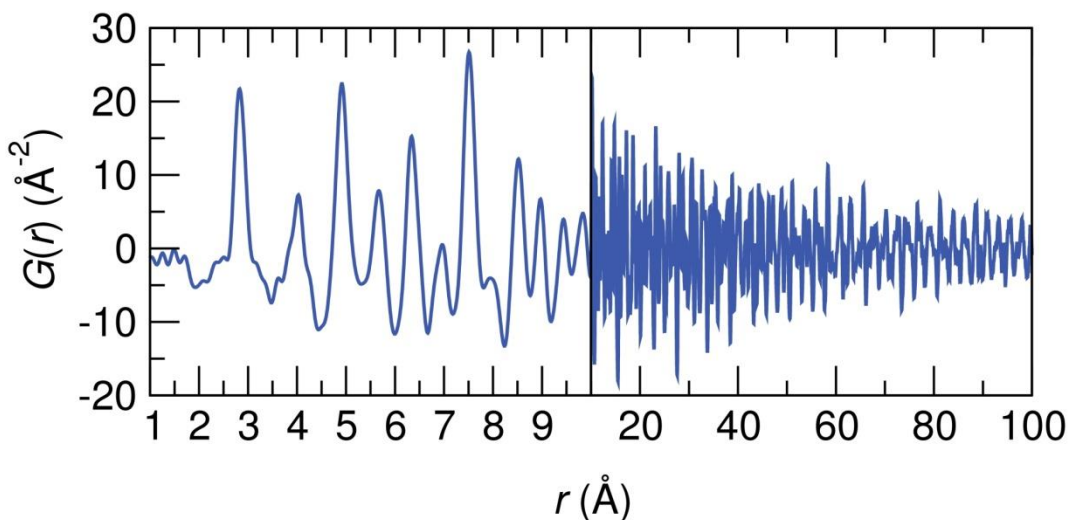
The Sine Fourier transform of the total (Bragg and diffuse) scattering



The total scattering structure factor: $S(Q)$



Sine Fourier transform



The Pair Distribution Function (PDF): $G(r)$

$$G(r) = \frac{2}{\pi} \int_{Q_{\min}}^{Q_{\max}} Q[S(Q) - 1] \sin(Qr) dQ$$

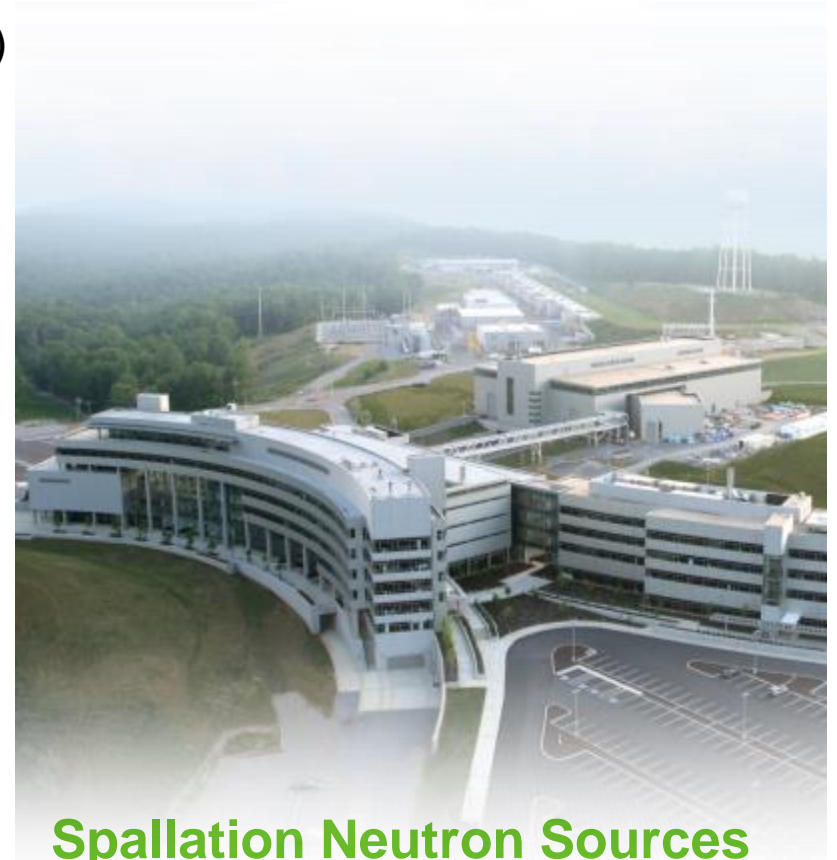
Obtaining High Quality PDFs

- (1) High maximum momentum transfer (Q_{\max})
- (2) Good Q-resolution, dQ
- (3) Good counting statistics
- (4) Low (and stable) instrument background

An ideal measurement would have no contribution from the instrument resolution

For PDF: a wide Q range and high flux is balanced with resolution

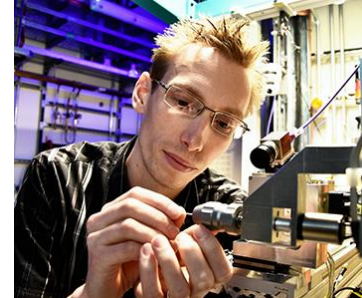
Synchrotron sources or
(high energy X-rays)



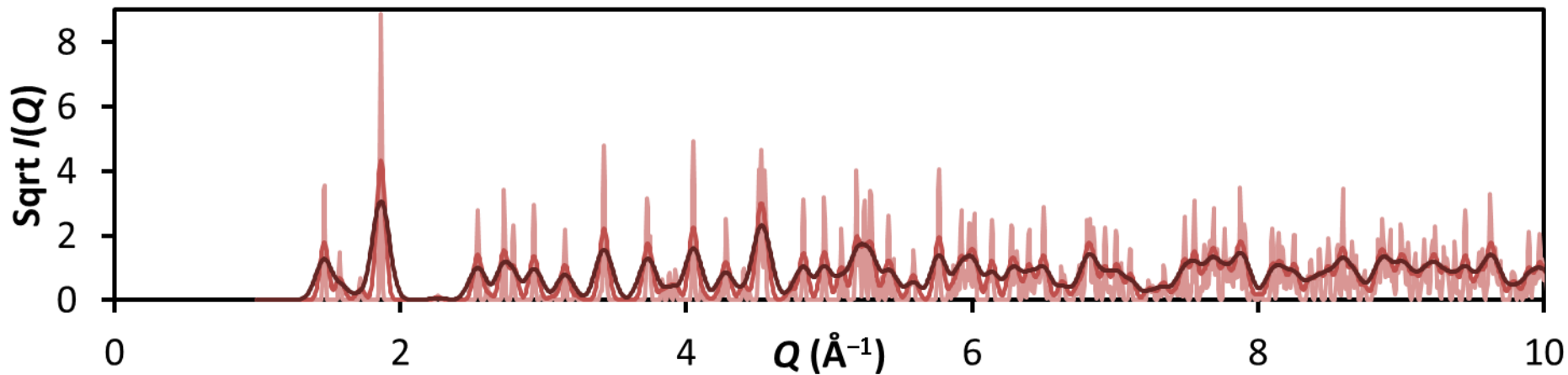
Spallation Neutron Sources
(reactor neutron energies are too low)

Resolution Effect

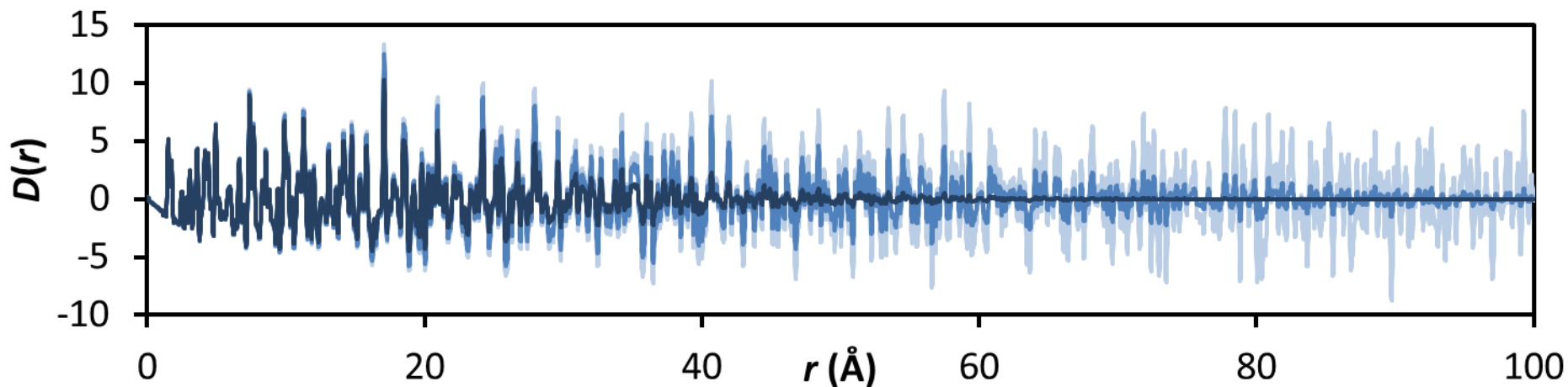
Courtesy of
Phil Chater,
Diamond Light
Source



Reciprocal space: Peak width, dQ



Real space: Damping with r



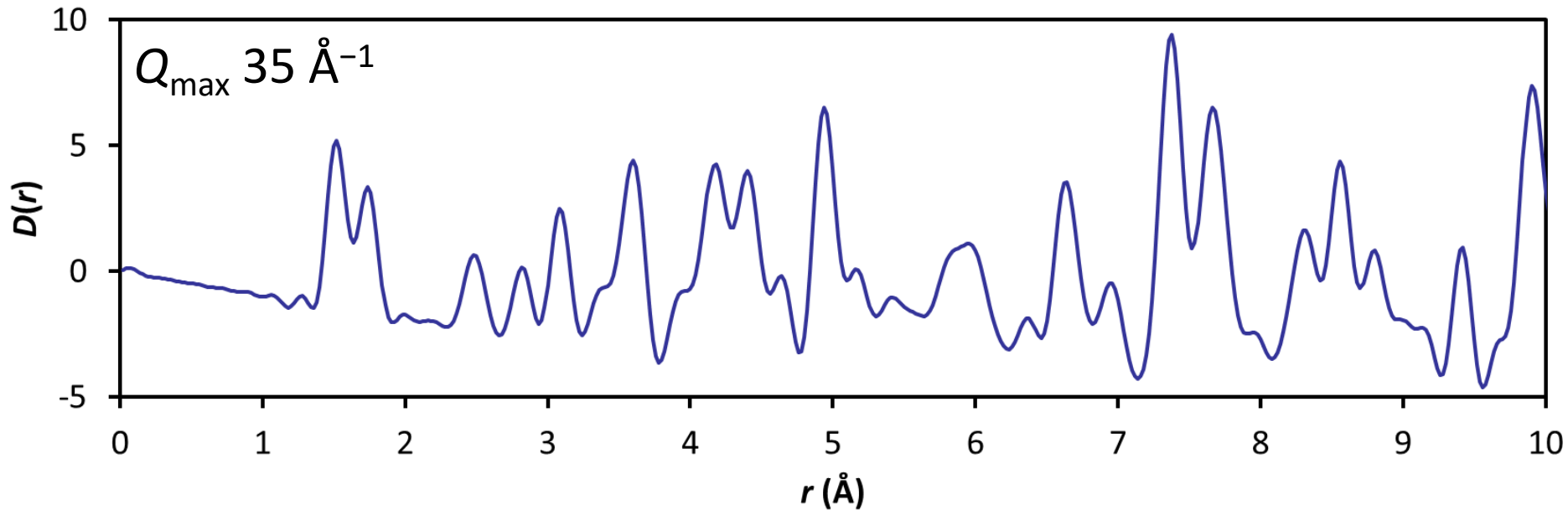
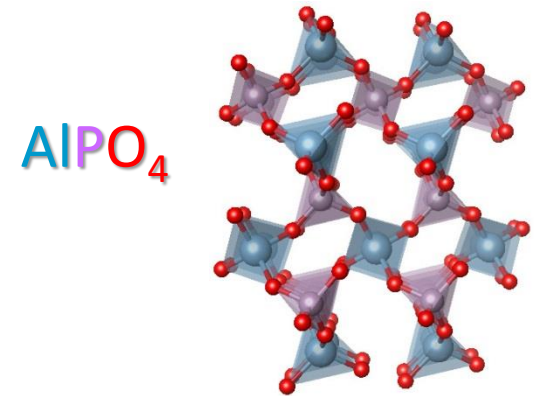
Q_{\max} Effect

Courtesy of
Phil Chater,
Diamond Light
Source

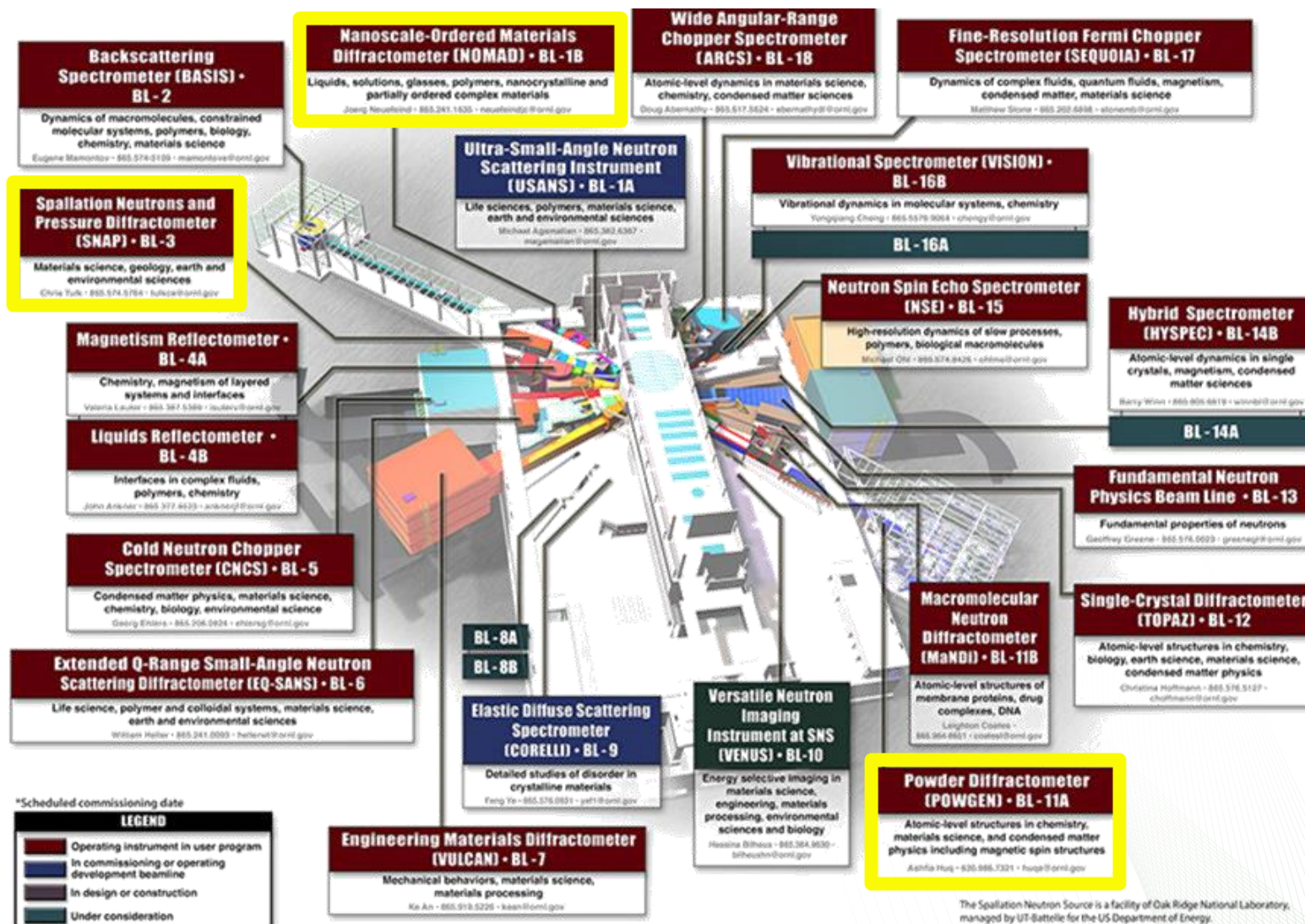


Δr resolution of a PDF is dominated by Q_{\max}

- $Q = 2\pi/d = 4\pi\sin\theta/\lambda$
- $\Delta r \approx 2\pi/Q_{\max}$



Instruments at SNS for PDF studies

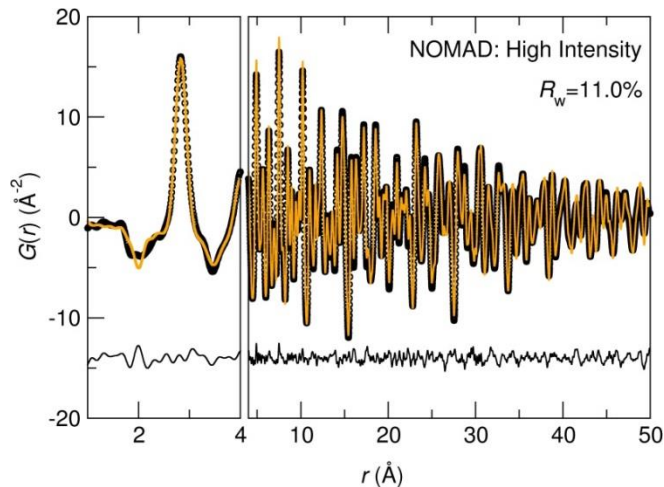


14-G00875A/gim

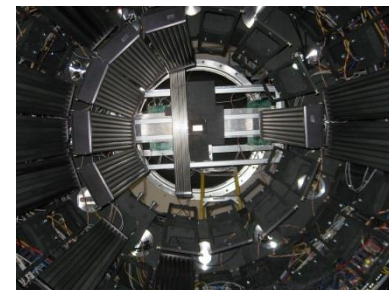
The Spallation Neutron Source is a facility of Oak Ridge National Laboratory, managed by UT-Battelle for the US Department of Energy.

Total Scattering at the SNS

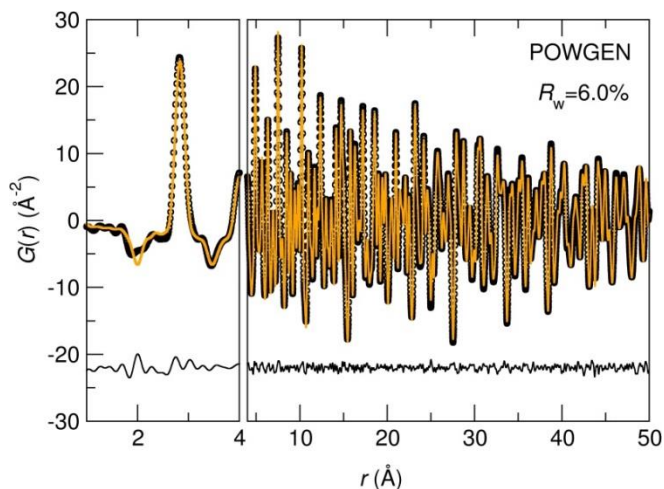
Pair Distribution Function (PDF) methods follow local atomic bonding configurations, intermediate structure, and correlation length scale- *regardless of a material's long range structure*



NOMAD data collected on **100 mg** of sample in a 3 mm quartz capillary in **3 hours**.



high intensity diffraction and PDF for small samples and *in situ* studies on amorphous, nanostructured, and crystalline materials



POWGEN data collected on **10 g** of sample in a 10 mm vanadium canister in **12 hours**.



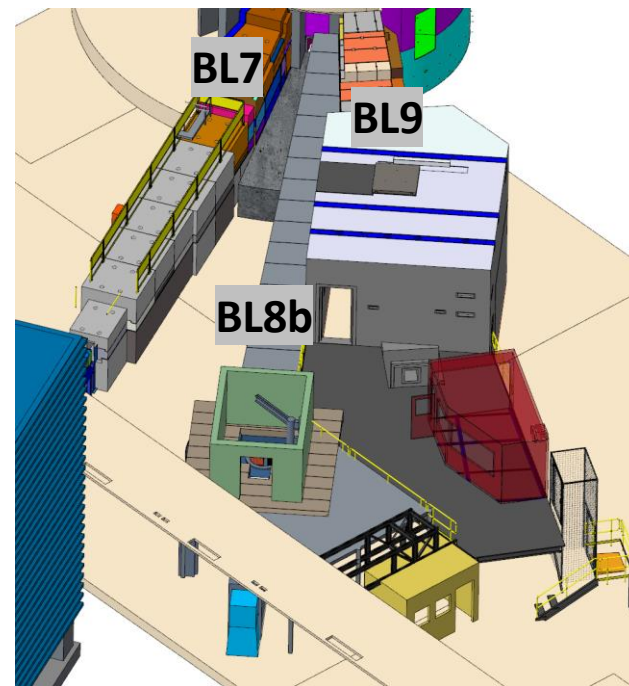
high resolution diffraction and PDF of crystalline materials

DISCOVER: Proposed Instrument for Diffraction and PDF, BL-8b

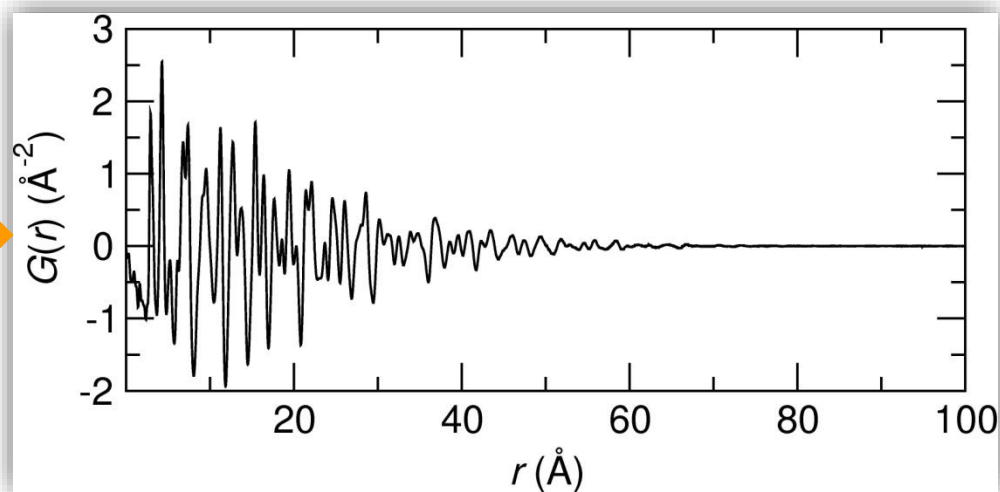
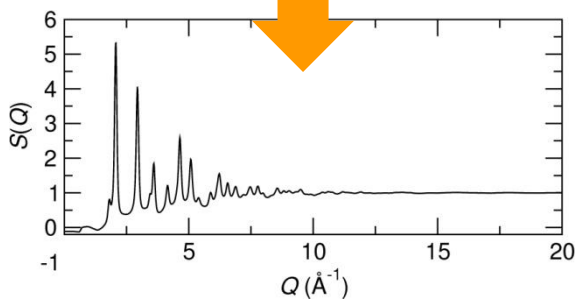
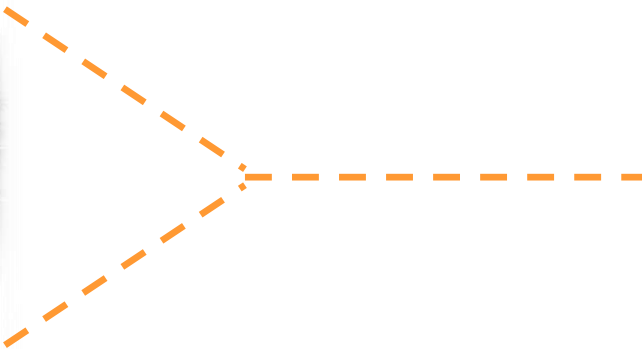
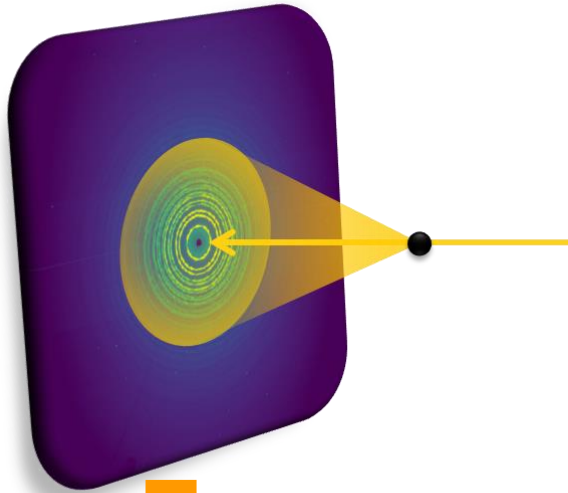
Materials Discovery

Understand the critical roles of heterogeneity, interfaces, and disorder

- What is the connection between global symmetry (i.e. that found over long length-scales) and local symmetry (i.e. that found at the atomic scale)?
- How does order evolve from the atomic to macroscale?
- Why does order evolve, and how can we control it?



Synchrotron Total Scattering: 2D Amorphous Si Detector



P. J. Chupas, K. W. Chapman, P. L. Lee, **Applications of an amorphous silicon-based area detector for high resolution, high sensitivity and fast time-resolved pair distribution function measurements**, *J. Appl. Crystallogr.* 40, 463, 2007.

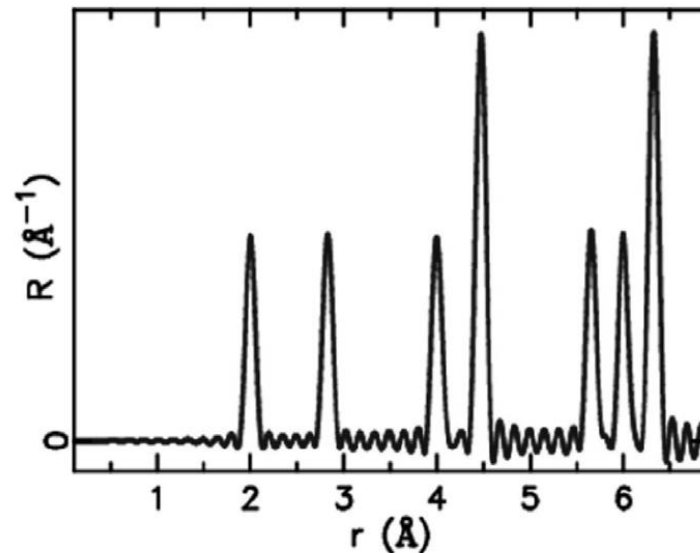
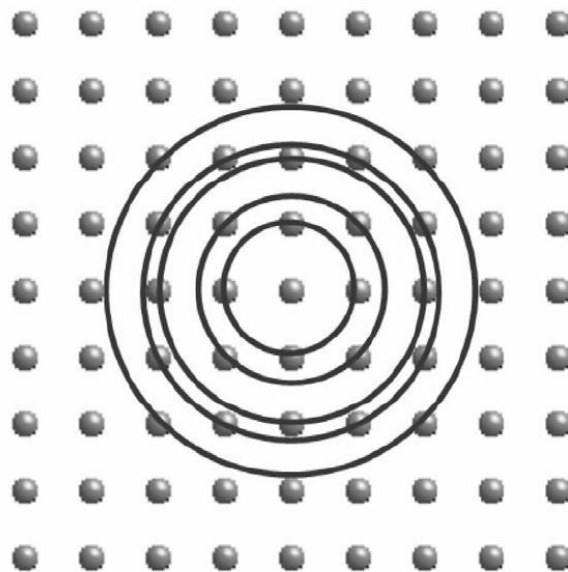
<http://dx.doi.org/10.1107/S0021889807007856>

Modeling a PDF



Pair Distribution Function

Based on the *radial distribution function* (RDF):



Atomic PDF (PDFFIT notation):

S.J.L Billinge, *Z. Kristallogr. Suppl.*,26,17 (2007)

$$G(r) = 4\pi r [\rho(r) - \rho_0]$$

atomic form factors
(Z for x-rays, b for neutrons)

sum over all atoms

$$G(r) = \sum_{ij} \left[\frac{b_i b_j}{\langle b \rangle^2} \delta(r - r_{ij}) \right] - 4\pi r \rho_0$$

distance between i and j atoms

average density

Calculating a PDF from a Model

Calculating a PDF from an atomistic model:

$$G(r) = \sum_{ij} \left[\frac{b_i b_j}{\langle b \rangle^2} \delta(r - r_{ij}) \right] - 4\pi r \rho_0$$

Peak Width

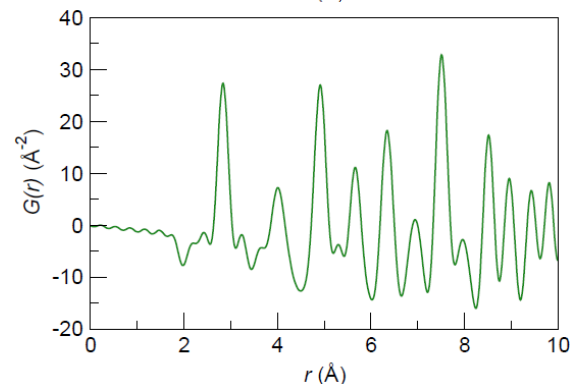
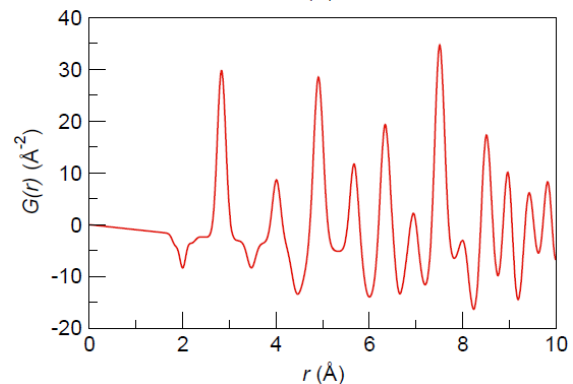
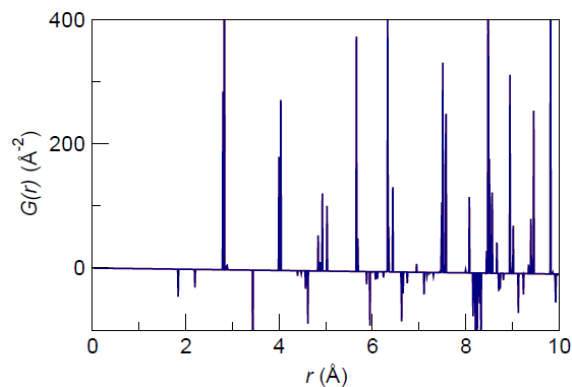
Small model: convolution of $\delta(r-r_{ij})$ with distribution function (*PDFgui* & *TOPAS v6*)

Large model: ensemble average of actual displacements (*RMCprofile*)

Termination ripples + instrumental dampening

Multiplication with step function in reciprocal space gives convolution with $\sin(Q_{\max}r)/r$ in real space

...



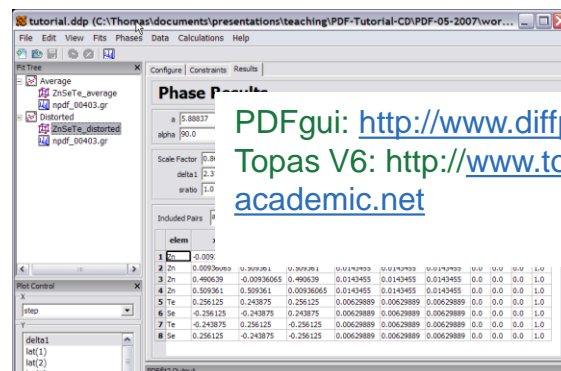
Atomic PDF Modeling

Small Models: Least Squares Refinement

Up to several hundred atoms

'Rietveld'-type parameters: *lattice parameters, atomic positions, displacement parameters, etc.*

Refinements as function of r -range



PDFgui: <http://www.diffpy.org/>
Topas V6: <http://www.topas-academic.net>

Large Model: Reverse Monte Carlo

20000 + atoms

Fit X-ray and neutron $F(Q)$, $G(r)$, Bragg profile

Constraints utilized

Static 3-D model of the structure (a snap-shot)

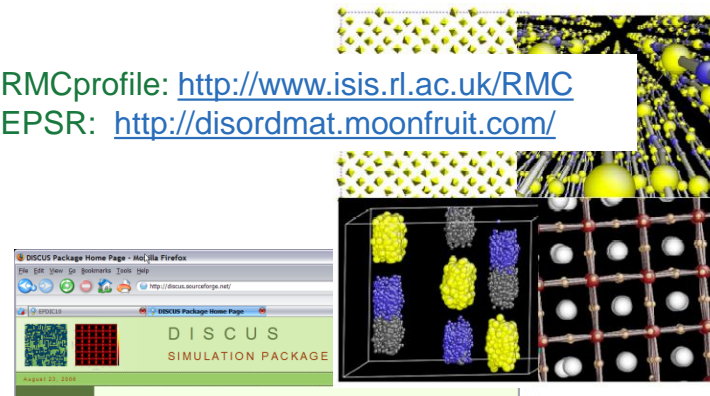
RMCprofile: <http://www.isis.rl.ac.uk/RMC>
EPSR: <http://disordmat.moonfruit.com/>

Multi-level / Complex Modeling

Refine higher level parameters (not each atom)

Example nanoparticle: *diameter, layer spacing, stacking fault probability*

Choose minimization scheme

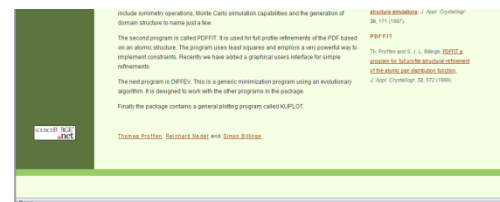


DIFFEV and DISCUS: <http://discus.sourceforge.net>
Topas V6: <http://www.topas-academic.net>

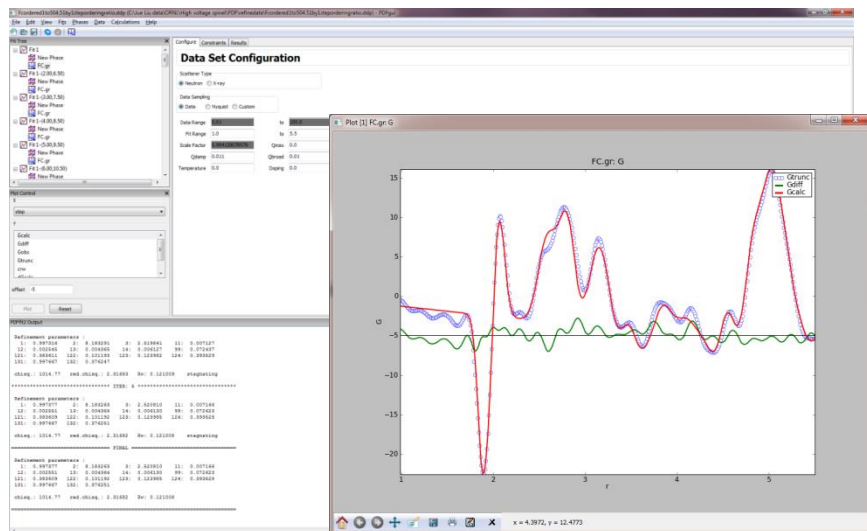
ab initio and force-field based approaches

Density Functional Theory

Molecular Dynamics



Small Box: Software Comparison

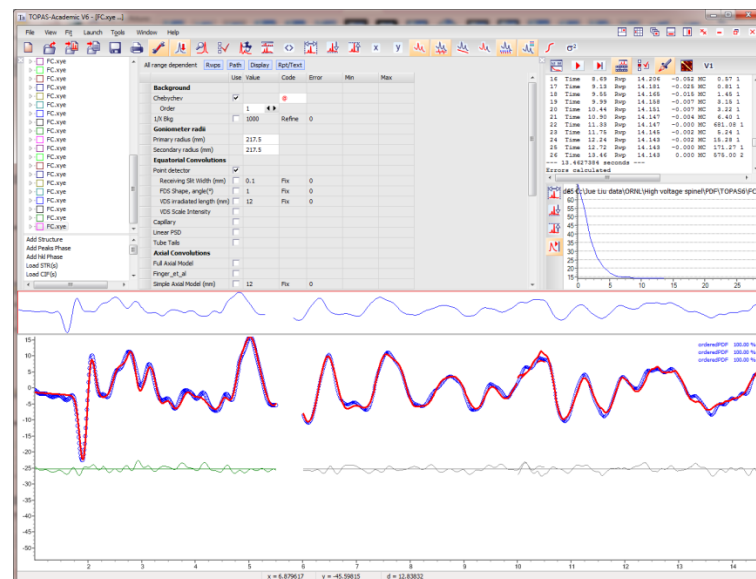


PDFgui <http://www.diffpy.org/>

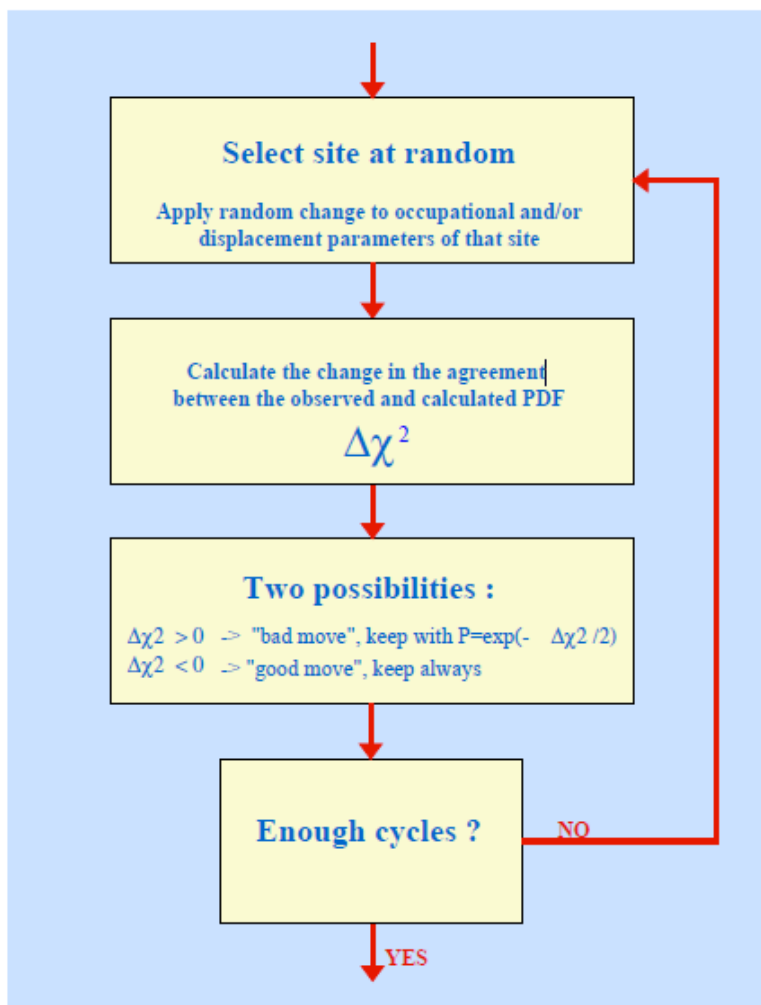
- + Open Source and Free
- + GUI is Simple and User-friendly
- Slow refinement, especially for high- r
- No longer actively developed

TOPAS PDF <http://www.topas-academic.net>

- + Super Fast
- + Flexible
- + Fit Bragg and PDF together
- Steeper learning curve
- Have to write your own macro



Large Box: Reverse Monte Carlo

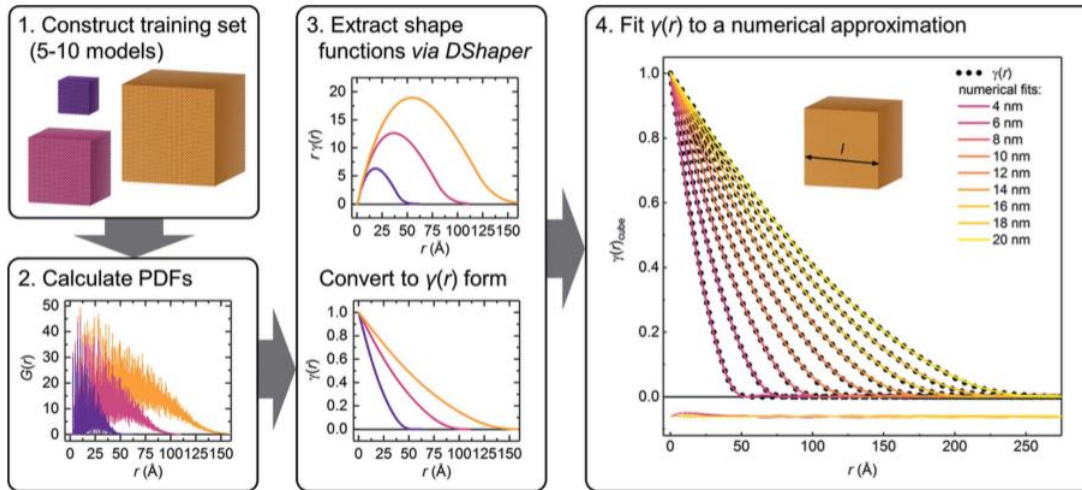


- **RMCprofile**

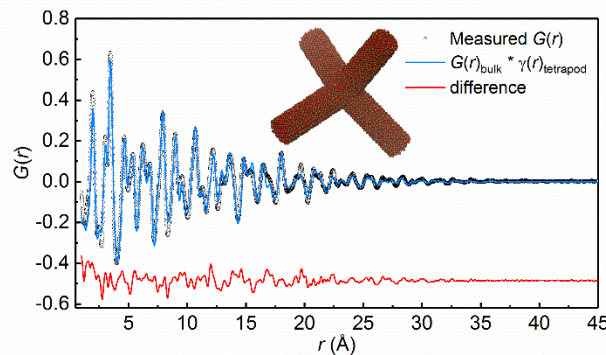
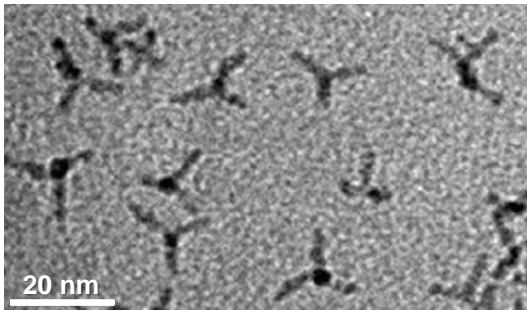
- Atomic configurations ~600 to 20000+ atoms
 - Fit both X-ray and neutron $F(Q)$
 - Fit $G(r)$
 - Fit Bragg profile (GSAS ToF 1-3)
 - Polyhedral restraints
 - Coordination constraints
 - Closest approach constraints
- **Produce a static 3-D model of the structure (a snap-shot in time)**
 - **Link:** <http://www.isis.rl.ac.uk/RMC>

Modeling nanoscale morphology in real space

$$G(r) = \frac{2}{\pi} \int_{Q_{min}}^{Q_{max}} Q[S(Q) - 1] \sin(Qr) dQ = 4\pi r [\rho(r) - \rho_0 \gamma_0(r)]$$



- $\gamma_0(r)$ is the particle shape function, it varies significantly from unity for nanomaterials and should be implemented as an r -dependent function
- Can fit physically-relevant shape parameters, such as a nanocube edge length, nanorod length and diameter, or arm length, width, and arm tip-to-arm tip distance in Fe_2O_3 tetrapods (left)



For use with Debye scattering approach: D. Olds, H.-W. Wang and K. Page, *J. Appl. Cryst.* **48**, 1651-1659 (2015).
 For use in small-box modeling approach: T.-M. Usher, D. Olds, J. Liu, K. Page, *Acta Cryst.* A74 (2018).

A Few Emerging Areas



Magnetic PDF: mPDF



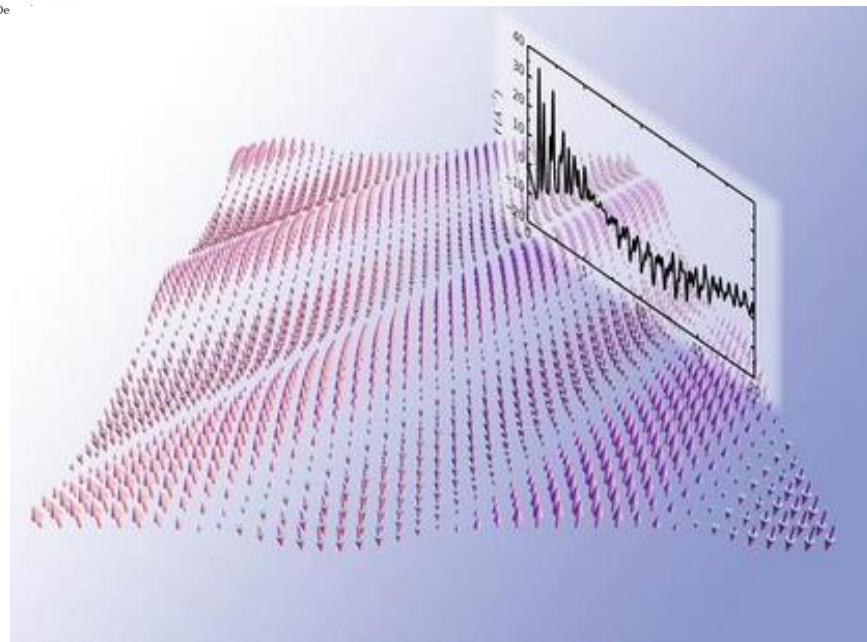
advances

Acta Crystallographica Section A
Foundations and
Advances
ISSN 2053-2733

Magnetic pair distribution function analysis of local magnetic correlations

Benjamin A. Frandsen,^a Xiaohao Yang^b and Simon J. L. Billinge^{b,c,*}

Received 10 October 2013
Accepted 6 December 2013



- Being developed to provide direct access to long-range and short-range magnetic correlations in real space
- **Spin order in diluted magnetic semiconductors, spin-stripe correlations in cuprate superconductors, spin fluctuations in frustrated magnetic systems, etc.**

ARTICLE

Received 14 Jul 2016 | Accepted 4 Nov 2016 | Published 20 Dec 2016

DOI: 10.1038/ncomms13842

OPEN

Emergent order in the kagome Ising magnet $\text{Dy}_3\text{Mg}_2\text{Sb}_3\text{O}_{14}$

Joseph A.M. Paddison^{1,2}, Harapan S. Ong¹, James O. Hamp¹, Paromita Mukherjee¹, Xiaojian Bai², Matthew G. Tucker^{3,4}, Nicholas P. Butch⁵, Claudio Castelnovo¹, Martin Mourigal² & S.E. Dutton¹

PRL 116, 197204 (2016)

PHYSICAL REVIEW LETTERS

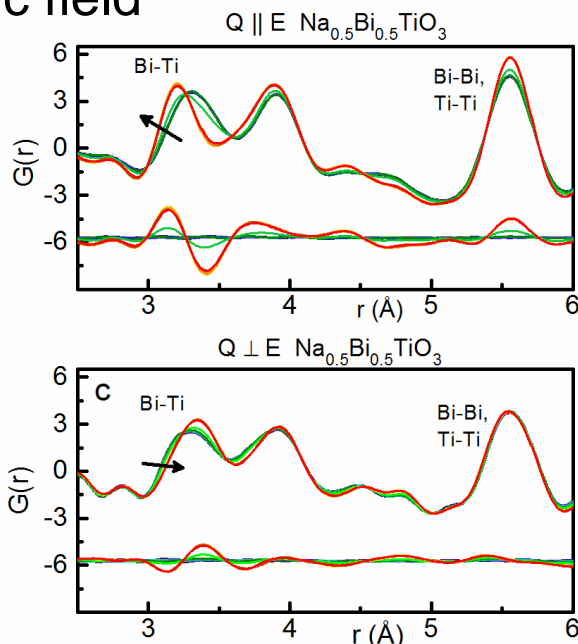
week ending
13 MAY 2016

Verification of Anderson Superexchange in MnO via Magnetic Pair Distribution Function Analysis and *ab initio* Theory

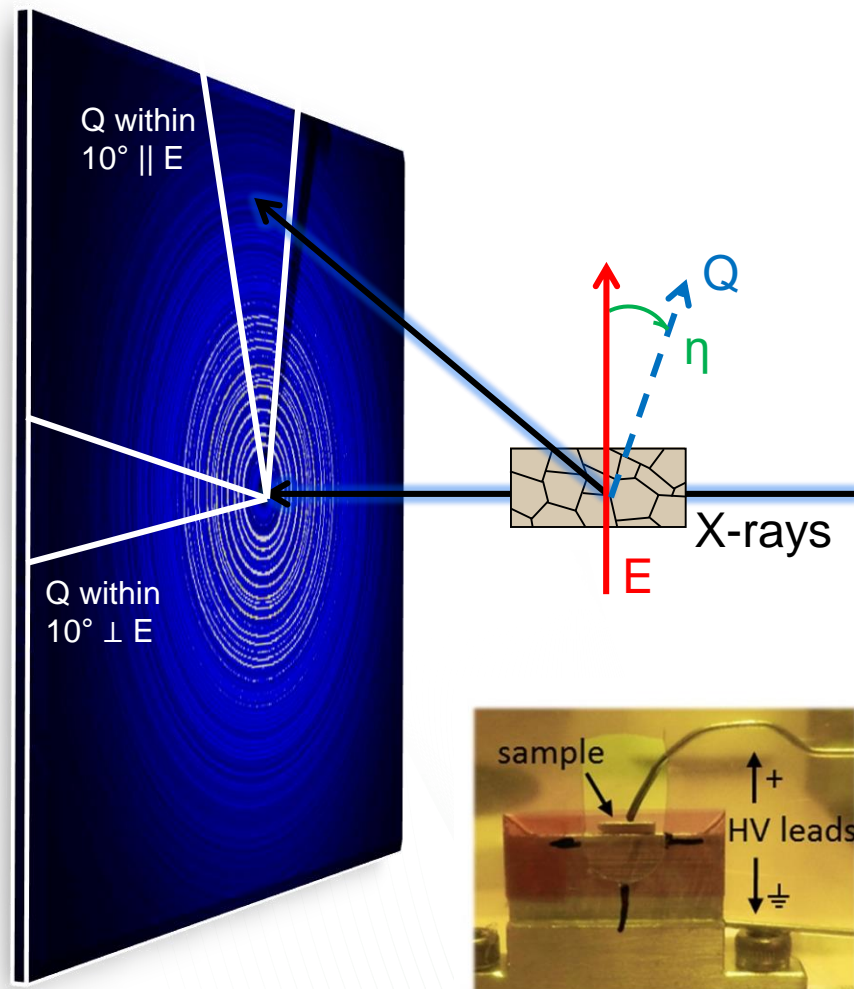
Benjamin A. Frandsen,¹ Michela Brunelli,² Katharine Page,³ Yasutomo J. Uemura,¹
Julie B. Staunton,⁴ and Simon J.L. Billinge^{5,6,*}

Field-Dependent PDF

- X-ray total scattering measured while static electric fields (0 to ~4 kV/mm) are applied to $\text{Na}_{1/2}\text{Bi}_{1/2}\text{TiO}_3$ polycrystalline ceramic samples
- Bi^{3+} reorientation observed at high electric field



11-ID-B, APS



T.-M. Usher, I. Levin, J.E. Daniels, and J.L. Jones, *Scientific Reports* 5, 14678 (2015).

A. J. Goetzee-Barral et al., *Phys. Rev. B* 96,, 014118 (2017).

When Should You Pursue PDF Studies of a Crystalline Material?

- ✓ You have modeled everything you can in reciprocal space
- ✓ You suspect the local structure may differ from the long-range structure

Why Would You Suspect a Distinct Local Structure?

Maybe...

- You find signatures of disorder through complementary methods
- An average structure model fails to explain observed material properties
- A theoretical study proposes an alternate structure to the one globally observed
- Lots of experience with a materials family or structural archetype

Some Resources and Programs

Data Collection

- Neutron: <http://neutronsources.org>
- X-ray: <http://www.lightsources.org>

Data Extraction

- PDFgetN: <http://pdfgetn.sourceforge.net>
- PDFgetX2/X3: <http://www.pa.msu.edu/cmp/billinge-group/programs/PDFgetX2/>
<http://www.diffpy.org/products/pdfgetx3.html>
- Gudrun: <http://disordmat.moonfruit.com/>
- ADDIE: ADvanced DIffraction Environment: coming soon from ORNL!

Data Modeling

- PDFgui: <http://www.diffpy.org/>
- Topas Academic: <http://www.topas-academic.net>
- RMCprofile: <http://www.isis.rl.ac.uk/RMC>
- DISCUS/DIFFEV: <http://discus.sourceforge.net>
- EPSR: <http://disordmat.moonfruit.com/>

References & Reviews

S.J.L. Billinge and I. Levin, The Problem with Determining Atomic Structure at the Nanoscale, *Science* 316, 561 (2007). <http://dx.doi.org/10.1126/science.1135080>

T. Egami and S. J. L. Billinge, *Underneath the Bragg peaks: structural analysis of complex materials*, Pergamon Press Elsevier, Oxford, England, 2003.

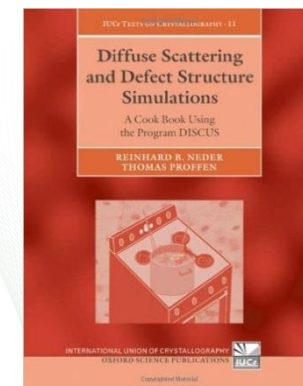
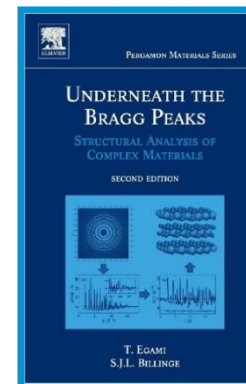
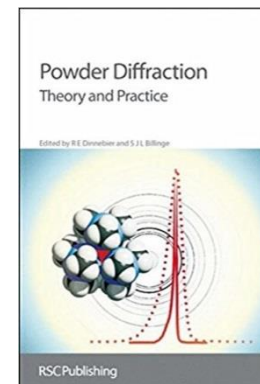
D. A. Keen, Derivation of commonly used functions for the pair distribution function, *technique J. Appl. Cryst.* 34 (2001) 172-177.

R. Neder and Th. Proffen, *Diffuse Scattering and Defect Structure Simulation*, Oxford University Press, 2008.

M. G. Tucker, M. T. Dove, and D. A. Keen, Application of the reverse Monte Carlo method to crystalline materials, *J. Appl. Cryst.* 34, 630-638 (2001).

D. A. Keen and A. L. Goodwin, The crystallography of correlated disorder, *Nature* 521, 303–309, 2015. <http://dx.doi.org/10.1038/nature14453>

H. Y. Playford, L. R. Owen, I. Levin, and M. G. Tucker, New insights into complex materials using Reverse Monte Carlo modeling, *Annual Review of Materials Research*, 44, 429-449, 2014. <http://dx.doi.org/10.1146/annurev-matsci-071312-121712>



Summary

Atomic PDF from total (Bragg and diffuse) scattering data gives access to:

Amorphous and nanomaterial structure

Departure from long range (average structure)

Displacements

Chemical short range order

Interstitials/vacancies

Correlation length scale of features (size)

Structure \leftrightarrow property relationships

Use multiple data sets (e.g. x-ray and neutron data, diffraction and PDF) to characterize complex materials

High-resolution instruments open the door to medium-range order investigations

Questions?



pagekl@ornl.gov